

# Coordination Compounds

## Examples of Multiple Choice Questions

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1. The \_\_\_\_\_ sphere is enclosed in brackets in formulas for complex species, and it includes the central metal ion plus the coordinated groups.  
(a) ligand  
(b) donor  
(c) oxidation  
(d) coordination  
(e) chelating
2. In coordination chemistry, the **donor atom** of a ligand is  
(a) a Lewis acid.  
(b) the counter ion  
(c) the central metal atom.  
(d) the atom in the ligand that shares an electron pair with the metal.  
(e) the atom in the ligand that accepts a share in an electron pair from the metal.
3. Consider the coordination compound,  $\text{Na}_2[\text{Pt}(\text{CN})_4]$ . The Lewis acid is  
(a)  $[\text{Pt}(\text{CN})_4]^{2-}$   
(b)  $\text{Na}^+$   
(c) Pt  
(d)  $\text{Pt}^{2+}$   
(e)  $\text{CN}^-$
4. Consider the coordination compound,  $\text{K}_2[\text{Cu}(\text{CN})_4]$ . A coordinate covalent bond exists between  
(a)  $\text{K}^+$  and  $\text{CN}^-$   
(b)  $\text{Cu}^{2+}$  and  $\text{CN}^-$   
(c)  $\text{K}^+$  and  $[\text{Cu}(\text{CN})_4]^{2-}$   
(d) C and N in  $\text{CN}^-$   
(e)  $\text{K}^+$  and  $\text{Cu}^{2+}$
- 5.

# Coordination Compounds Multiple Choice Questions

**John T. Moore, Richard H. Langley**



## Coordination Compounds Multiple Choice Questions:

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## **Table of Contents Coordination Compounds Multiple Choice Questions**

1. Understanding the eBook Coordination Compounds Multiple Choice Questions
  - The Rise of Digital Reading Coordination Compounds Multiple Choice Questions
  - Advantages of eBooks Over Traditional Books
2. Identifying Coordination Compounds Multiple Choice Questions
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Coordination Compounds Multiple Choice Questions
  - User-Friendly Interface
4. Exploring eBook Recommendations from Coordination Compounds Multiple Choice Questions
  - Personalized Recommendations
  - Coordination Compounds Multiple Choice Questions User Reviews and Ratings
  - Coordination Compounds Multiple Choice Questions and Bestseller Lists
5. Accessing Coordination Compounds Multiple Choice Questions Free and Paid eBooks
  - Coordination Compounds Multiple Choice Questions Public Domain eBooks
  - Coordination Compounds Multiple Choice Questions eBook Subscription Services
  - Coordination Compounds Multiple Choice Questions Budget-Friendly Options
6. Navigating Coordination Compounds Multiple Choice Questions eBook Formats
  - ePub, PDF, MOBI, and More
  - Coordination Compounds Multiple Choice Questions Compatibility with Devices
  - Coordination Compounds Multiple Choice Questions Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Coordination Compounds Multiple Choice Questions
  - Highlighting and Note-Taking Coordination Compounds Multiple Choice Questions
  - Interactive Elements Coordination Compounds Multiple Choice Questions



8. Staying Engaged with Coordination Compounds Multiple Choice Questions
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Coordination Compounds Multiple Choice Questions
9. Balancing eBooks and Physical Books Coordination Compounds Multiple Choice Questions
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Coordination Compounds Multiple Choice Questions
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Coordination Compounds Multiple Choice Questions
  - Setting Reading Goals Coordination Compounds Multiple Choice Questions
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Coordination Compounds Multiple Choice Questions
  - Fact-Checking eBook Content of Coordination Compounds Multiple Choice Questions
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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