

Covalent Bonding Worksheet

Covalent bonding occurs when two or more NON_METALS share electrons, attempting to attain a stable octet (8 outer electrons) in their outer shell for at least part of the time. Draw a Lewis dot diagram for each element listed. Circle the unpaired electrons that will be shared between the elements.

1.) H_2 hydrogen is diatomic:



2.) F_2 fluorine is diatomic:



3.) O_2 oxygen is diatomic:



4.) N_2 nitrogen is diatomic. Is this a triple bond?



5.) BF_3 you need 3 fluorine atoms here:



6.) Ammonia NH_3 hint: how many hydrogen atoms are needed?



7.) Carbon dioxide CO_2



8.) Methane CH_4 careful here 4 hydrogen atoms needed



9.) Dihydrogen monoxide: the most dangerous substance on the planet. It has killed more people than any other substance known to mankind!!



10.) SO_2 hint: one pair of electrons from sulfur must be slit up for this one to work.



Covalent Bonding Workbook Answers

JR Anderson



Covalent Bonding Workbook Answers:

Adopting the Track of Expression: An Mental Symphony within **Covalent Bonding Workbook Answers**

In a global used by screens and the ceaseless chatter of fast conversation, the melodic splendor and mental symphony developed by the written word usually diminish into the backdrop, eclipsed by the persistent sound and distractions that permeate our lives. Nevertheless, located within the pages of **Covalent Bonding Workbook Answers** a charming fictional treasure full of organic emotions, lies an immersive symphony waiting to be embraced. Crafted by a masterful composer of language, that fascinating masterpiece conducts visitors on a mental journey, well unraveling the hidden songs and profound affect resonating within each carefully crafted phrase. Within the depths of the moving analysis, we will discover the book is main harmonies, analyze its enthralling writing fashion, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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