

Covalent Bonding & Molecular Compounds Multiple Choice Review

PSI Chemistry

Name _____

1) Which pair of elements is most apt to form a molecular compound with each other?

- A) aluminum, oxygen
- B) magnesium, iodine
- C) sulfur, fluorine
- D) potassium, lithium
- E) barium, bromine

2) Which of the following bonds would be best categorized as covalent?

- I. H-S
- II. Al-S
- III. N-F

- A) I only
- B) II only
- C) III only
- D) I and III
- E) I, II, and III

3) Which of the following BEST describes the bonding found within solid Al_2O_3 ?

- A) Strong covalent bonds between atoms with similar electronegativities
- B) Covalently bound atoms arranged in small individual molecules.
- C) Electrostatic attractions between + and - charged ions
- D) Positively charged ions covalently bound with many mobile electrons
- E) None of these

4) Which of the following species below would be considered molecular in nature?

- A) C(diamond)
- B) C(graphite)
- C) Fe
- D) AlCl_3
- E) PCl_5

5) The substance below BEST characterized as having a high melting point and able to conduct electricity in the liquid state only would be:

- A) CH_4
- B) V_2O_5
- C) CO
- D) HF
- E) C(diamond)

6) A material melts at -77°C , is non-conductive, and consists of small individual molecules held together by inter-molecular forces. Which of the molecules below is this material likely to be?

- A) NaCl

Covalent Bonding Molecular Compounds Multiple Choice

Susan McMurry



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