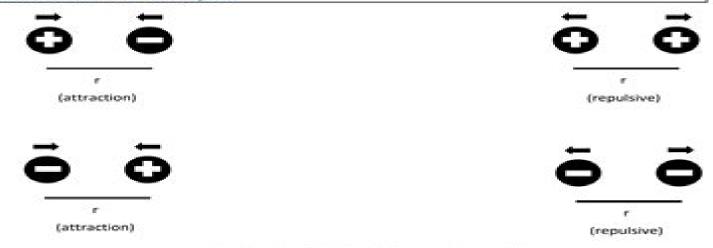
# Coulomb's Law

This law is called Coulomb's law because it comes from a French scientist named Charles Augustin Coulomb (1736-1806). This scientist investigates the repulsive or attractive forces of two electrically charged objects with respect to the magnitude of the electric charge and their distance using a torsion balance. Based on experiments using a torsion balance, Coulomb concluded the interaction of two charged objects as follows:

- The greater the distance of the two objects charged, the smaller the electric force between the two objects and vice versa.
- The greater the charge of the two objects, the greater the electric force between the objects.

The attraction force or refuse to reject the two electrically charged objects is directly proportional (comparable) with the charge of each object and inversely proportional to the square of the distance between the two objects.



The formula of Coulomb's law mathematically:

$$F_c = k \frac{q_1 q_2}{r^2}$$

Information:

F<sub>c</sub> = Coulomb force (N)

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