

Counting Atoms Sheet - Answers

Name each of the following chemical compounds (make sure you use the right naming scheme!) and determine how many atoms of each element are present:

- 1) CaF_2 calcium fluoride, 1 calcium, 2 fluorine
- 2) $\text{Be}(\text{OH})_2$ beryllium hydroxide, 1 Be, 2 O, 2 H
- 3) NO_2 nitrogen dioxide, 1 nitrogen, 2 oxygen
- 4) $\text{Al}_2(\text{SO}_4)_3$ aluminum sulfate, 2 Al, 3 S, 12 O
- 5) NH_4NO_3 ammonium nitrate, 2 N, 4 H, 3 O
- 6) S_2F_2 disulfur difluoride, 2 sulfur, 2 fluorine
- 7) Na_2CO_3 sodium carbonate, 2 Na, 1 C, 3 O
- 8) CH_4 methane, 1 carbon, 4 hydrogen

Write the formulas for each of the following chemical compounds (make sure you use the right naming scheme!) and determine how many atoms of each element are present:

- 9) phosphorus trichloride PCl_3 , 1 phosphorus, 3 chlorine
- 10) magnesium hydroxide $\text{Mg}(\text{OH})_2$, 1 Mg, 2 O, 2 H
- 11) potassium phosphate K_3PO_4 , 3 K, 1 P, 4 O
- 12) diphosphorus tetrabromide P_2Br_4 , 2 phosphorus, 4 bromine
- 13) ammonia NH_3 , 1 nitrogen, 3 hydrogen
- 14) germanium phosphate $\text{Ge}_3(\text{PO}_4)_4$, 3 Ge, 4 P, 16 O
- 15) ammonium sulfate $(\text{NH}_4)_2\text{SO}_4$, 2 N, 8 H, 1 S, 4 O
- 16) diphosphorus pentoxide P_2O_5 , 2 P, 5 O

Counting Atoms Answers

Paul A. Tipler, Gene Mosca



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