

1. A 33.7-g sample of HF is dissolved in water to give  $2.0 \times 10^2$  mL of solution. The concentration of the solution is:

A) 1.68 M  
B) 0.34 M  
C) 0.17 M  
**D) 8.4 M**  
E) 16.9 M

$$[HF] = \frac{33.7g HF}{2.0 \times 10^2 mL} \times \frac{1000 mL}{1 L} \times \frac{1 mol HF}{20.01g HF} = 8.4 M$$

2. What mass of calcium chloride,  $CaCl_2$ , is needed to prepare 3.419 L of a 1.56 M solution?

A) 243 g  
B) 5.33 g  
C) 50.6 g  
D) 111 g  
**E) 592 g**

$$\frac{3.419 L}{1} \times \frac{1.56 mol CaCl_2}{L} \times \frac{110.98g CaCl_2}{1 mol CaCl_2} = 592g CaCl_2$$

3. A 51.24-g sample of  $Ba(OH)_2$  is dissolved in enough water to make 1.205 liters of solution. How many mL of this solution must be diluted with water in order to make 1.000 liter of 0.1000 molar  $Ba(OH)_2$ ?

**A) 403 mL**  
B) 248 mL  
C) 24.8 mL  
D) 83.0 mL  
E) 331 mL

$$[Ba(OH)_2] = \frac{51.24g Ba(OH)_2}{1.205 L} \times \frac{1 mol Ba(OH)_2}{171.36g Ba(OH)_2} = 0.2481 M$$

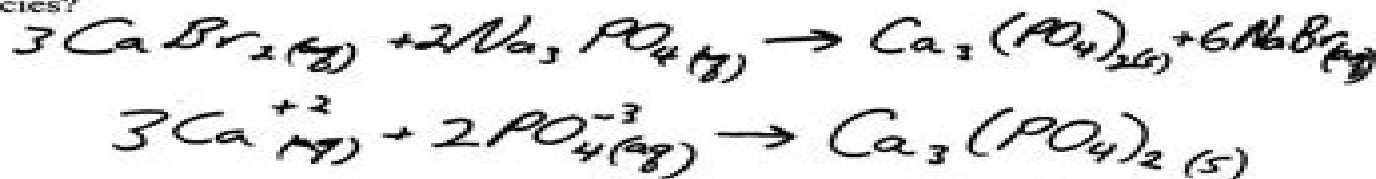
$$M_1 = 0.1000 M \quad M_2 = 0.2481 M$$

$$V_1 = 1.000 L \quad V_2 = ?$$

$$V_2 = \frac{0.1000 M \times 1.000 L}{0.2481 M} = 0.4031 L = 403 mL$$

4. The net ionic equation for the reaction of calcium bromide and sodium phosphate contains which of the following species?

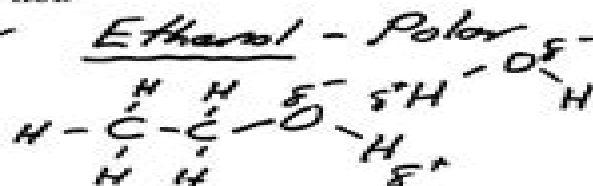
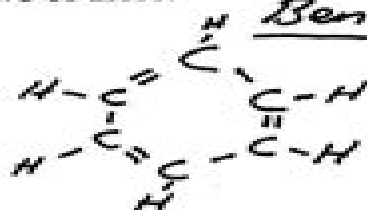
A)  $Ca^{2+}(aq)$   
B)  $PO_4^{3-}(aq)$   
C)  $2Ca_3(PO_4)_2(s)$   
D)  $6NaBr(aq)$   
**E)  $3Ca^{2+}(aq)$**



5. Consider two organic molecules, ethanol and benzene. One dissolves in water and the other does not. Why?

A) They have different molar masses.  
B) One is ionic, the other is not.  
C) One is an electrolyte, the other is not.  
**D) Ethanol contains a polar O—H bond, and benzene does not.**  
E) Two of these.

\* Know the general solubility rule of "Likes Dissolve Likes"



# High School Chemistry Test Questions And Answers

**Marvin James Van Wagenen**



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learning because the tools and strategies she shares have enabled faculty to facilitate dramatic improvements in student learning and success This book encapsulates the model and ideas she has developed in the past fifteen years ideas that are being adopted by an increasing number of faculty with considerable effect The methods she proposes do not require restructuring courses or an inordinate amount of time to teach They can often be accomplished in a single session transforming students from memorizers and regurgitators to students who begin to think critically and take responsibility for their own learning Saundra McGuire takes the reader sequentially through the ideas and strategies that students need to understand and implement First she demonstrates how introducing students to metacognition and Bloom s Taxonomy reveals to them the importance of understanding how they learn and provides the lens through which they can view learning activities and measure their intellectual growth Next she presents a specific study system that can quickly empower students to maximize their learning Then she addresses the importance of dealing with emotion attitudes and motivation by suggesting ways to change students mindsets about ability and by providing a range of strategies to boost motivation and learning finally she offers guidance to faculty on partnering with campus learning centers She pays particular attention to academically unprepared students noting that the strategies she offers for this particular population are equally beneficial for all students While stressing that there are many ways to teach effectively and that readers can be flexible in picking and choosing among the strategies she presents Saundra McGuire offers the reader a step by step process for delivering the key messages of the book to students in as little as 50 minutes Free online supplements provide three slide sets and a sample video lecture This book is written primarily for faculty but will be equally useful for TAs tutors and learning center professionals For readers with no background in education or cognitive psychology the book avoids jargon and esoteric theory     **Journal of Education** ,1914     **The High School Teacher** ,1925

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