

# BEST YOUTUBE SHORTS IDEAS



# Proven Short Form Content Ideas In 2025

**Jesse Lane Shepherd**



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**Public Relations and Death** Roumen Dimitrov, Jordi Xifra, 2025-12-29 This volume for the first time examines the intricate and often obscured interplay between death and public relations across historical periods and within contemporary society Beyond its natural inevitability death has functioned as a potent instrument for image and reputation management the legitimisation of authority and the construction of collective memory From antiquity to the digital era communicative practices have employed death not merely to commemorate the deceased but also as a mechanism to consolidate dynastic power disseminate ideological frameworks and exert influence over the living Relations to death and relations through death shape the society s past present and future This generative study explores the tensions between individual memory and official historiography the strategic use of selective commemoration to legitimate political and cultural power and the role of public relations in shaping narratives of life and death across the arts media and political discourse This comprehensive volume will be of interest to scholars and researchers of public relations communication studies promotional marketing and public diplomacy as well as critical historiography thanatology sociology political philosophy garbology and discursive approaches to age and death

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web in the context of a chemical reaction the term neutralization is used for a reaction between an acid and a base or alkali historically this reaction was represented as acid base alkali salt water for example  $\text{hcl naoh nacl h}_2\text{o}$

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web during this reaction the  $\text{h}^+$  ions from the acid and the  $\text{oh}^-$  ions from the alkali react together to form water as this happens energy is released which is detected by the thermometer

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web for reactions involving acetic acid or ammonia the measured enthalpy change of neutralization is a few kJ less exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of sodium hydroxide solution with HCl as 57.9 kJ mol<sup>-1</sup>

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web sep 20 2014 i will illustrate this with some examples when an acid is neutralised by an alkali the reaction is exothermic eg 1 HCl(aq) + NaOH(aq) → NaCl(aq) + H<sub>2</sub>O(l) for which ΔH = 57 kJ mol<sup>-1</sup> eg 2 HNO<sub>3</sub>(aq) + KOH(aq) → KNO<sub>3</sub>(aq) + H<sub>2</sub>O(l) for which ΔH = 57 kJ mol<sup>-1</sup>

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