

Acids and Bases Worksheet

1. Acids and bases share some properties but are different in others. Use the samples at the side of the class to complete the chart. If both acids and bases share the property, place it in both columns.

Property	Acids	Bases
Conductivity in Aqueous Solution	very	very
Taste	Sour	Bitter
Feel	—	Slippery
Reaction with Metal	produces H_2 gas	—
Reaction with Carbonate	produces CO_2 gas	—
Colour on Litmus Paper	Red	Blue
Colour with Universal Indicator	Red	Purple
Colour with phenolphthalein	Colourless	pink

2. Indicate if each compound is an acid or base and name it.

	Acid/Base?	Name
HNO_3	<u>A</u>	<u>nitric acid</u>
$Mg(OH)_2$	<u>B</u>	<u>magnesium hydroxide</u>
H_3PO_4	<u>A</u>	<u>phosphoric acid</u>
H_2CO_3	<u>A</u>	<u>carbonic acid</u>
HF	<u>A</u>	<u>hydrofluoric acid</u>
$Fe(OH)_3$	<u>B</u>	<u>iron (iii) hydroxide</u>

3. Write the formula for each of the following acids/bases:

hydrochloric acid	<u>HCl</u>	hydrochloric acid	<u>HCl</u>
aluminum hydroxide	<u>$Al(OH)_3$</u>	sodium hydroxide	<u>NaOH</u>
lithium hydroxide	<u>LiOH</u>	calcium hydroxide	<u>$Ca(OH)_2$</u>
sulfuric acid	<u>H_2SO_4</u>	sulfurous acid	<u>H_2SO_3</u>

Acids Bases Chemistry Test Answers

John Ferguson Snell



Acids Bases Chemistry Test Answers:

Acids, Bases, and Solutions Elizabeth R. Kellerman, 1999-09-01 The activities in this book explain elementary concepts in the study of chemistry including acids bases solvents solutions crystals and more General background information suggested activities questions for discussion and answers are included

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1. Understanding the eBook Acids Bases Chemistry Test Answers
 - The Rise of Digital Reading Acids Bases Chemistry Test Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Acids Bases Chemistry Test Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Acids Bases Chemistry Test Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Acids Bases Chemistry Test Answers
 - Personalized Recommendations
 - Acids Bases Chemistry Test Answers User Reviews and Ratings
 - Acids Bases Chemistry Test Answers and Bestseller Lists
5. Accessing Acids Bases Chemistry Test Answers Free and Paid eBooks
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- Acids Bases Chemistry Test Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Acids Bases Chemistry Test Answers
 - Highlighting and Note-Taking Acids Bases Chemistry Test Answers
 - Interactive Elements Acids Bases Chemistry Test Answers
- 8. Staying Engaged with Acids Bases Chemistry Test Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Acids Bases Chemistry Test Answers
- 9. Balancing eBooks and Physical Books Acids Bases Chemistry Test Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Acids Bases Chemistry Test Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Acids Bases Chemistry Test Answers
 - Setting Reading Goals Acids Bases Chemistry Test Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Acids Bases Chemistry Test Answers
 - Fact-Checking eBook Content of Acids Bases Chemistry Test Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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