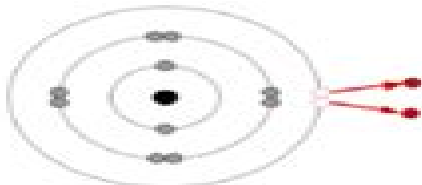
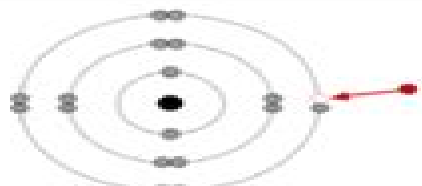


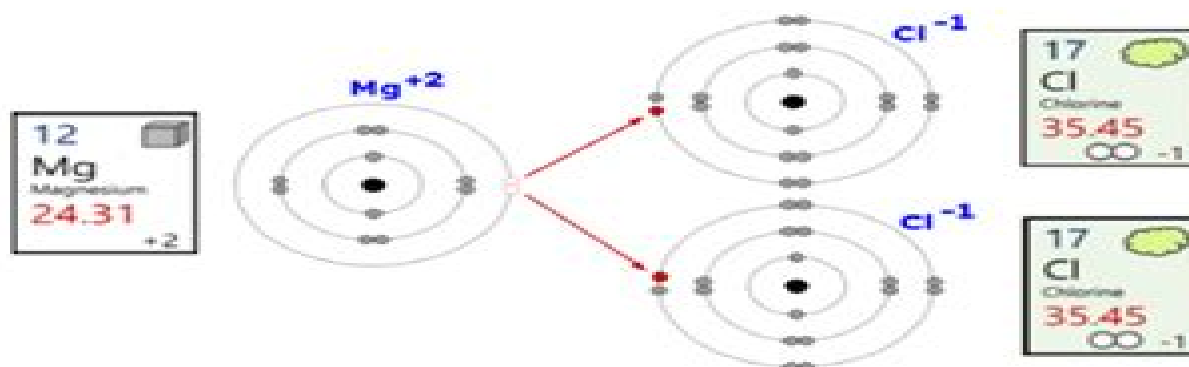
# Ionic Bonding Puzzle Activity

Name: \_\_\_\_\_

When atoms of different elements bond together, they produce a compound. One category of compounds, known as **Ionic Compounds**, is formed when atoms transfer electrons to other atoms and become charged. The charged atoms are called **ions** and must combine in the correct ratio so that the overall charge is neutral (charge = 0).

<div><div>12</div><div>Mg</div><div>Magnesium</div><div>24.31</div><div>+2</div></div>		Magnesium starts with 12 electrons and then <u>loses</u> 2 to make a complete outer energy shell	<div>Protons = 12</div> <div>Electrons = 10</div> <div>Charge = +2</div>
<div><div>17</div><div>Cl</div><div>Chlorine</div><div>35.45</div><div>-1</div></div>		Chlorine starts with 17 electrons and then <u>gains</u> 1 to make a complete outer energy shell	<div>Protons = 17</div> <div>Electrons = 18</div> <div>Charge = -1</div>

**Example:** During a chemical reaction between Magnesium and Chlorine, the following occurs:



*This creates the neutral compound called Magnesium chloride. 1 atom of magnesium combines with 2 atoms of chlorine with the chemical formula  $MgCl_2$ .*

# Activity On Ionic Bonding With Answers Thepumpore

**D Kirk**



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