AP Biology Lab 2 Enzyme Catalysis

Introduction.

Enzymes catalyze reactions by lowering the activation energy necessary for a reaction to occur.



In this taboratory, you will study some of the basic principles of molecuar movement in solution and perform a series of activities to investigate these processes:

Key Concepts

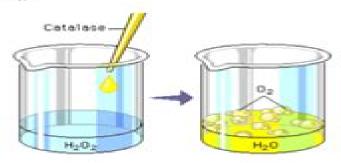
Encrymen catalyce reactions by lowering the activation energy necessary for a reaction to occur. The molecule that an encryme acts on is called the substrate. In an encryme-mediated reaction, substrate molecules are changed, and product is formed. The encryme molecule is unchanged after the reaction, and it can continue to catalyze the same type of reaction over and over.

Each enzyme is specific for the reaction it will catalyze. In this laboratory,

Emzyme = catalase Substrate = hydrogen peroxide (HuO₂) Products = water and oxygen

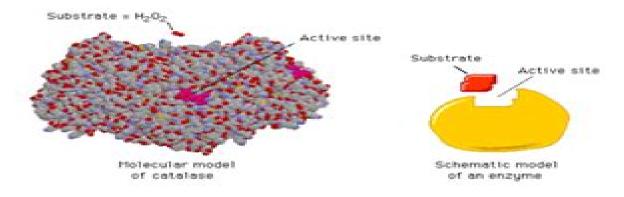
If a small amount of catalase is added to hydrogen peroxide, you will be able to observe bubbles of oxygen forming.

Over the next several screens we will take a closer look at what happens during an enzymatic reaction. We begin with a model of enzyme structure.



Enzyme Structure

Enzymes are globular proteins. Their folded conformation creates an area known as the active site. The nature and arrangement of amino acids in the active site make it specific for only one type of substrate.



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Cathrine M. Monson

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Aqueous-phase Catalytic Conversions of Renewable Feedstocks for Sustainable **Fulltext Sources Online** .2007 Biorefineries Georgios Papadogianakis, Roger Arthur Sheldon, Yulong Wu, Dmitry Yu. Murzin, 2024-08-23 Today there is growing interest in aqueous phase catalytic conversions for the valorization of renewable biomass based feedstocks for biorefineries to produce in a sustainable way biofuels chemicals power energy materials pharmaceuticals and food This is because of the highly polar nature of water which makes it an ideal medium to convert polar biomass based lignocellulose cellulose hemicellulose lignin with high oxygen content and their upgraded products such as hydrophilic carbohydrates platform chemicals and their derivatives Another reason which makes water the solvent of choice is that water itself is involved either as a reagent or as a byproduct even in large amounts in typical conversions for the valorization of biomass The obtained intermediates further react in the aqueous medium often without any separation and purification to manufacture more valuable products This results in substantial energy savings lower emissions and economic benefits Furthermore water could act as a catalyst in conversions of biomass based feedstocks such as in liquefaction reactions under subcritical conditions Moreover novel types of catalytic reactivity have been observed in the aqueous solvent not only with water soluble transition metal catalytic complexes but also with conventional heterogeneous catalysts and catalytic nanoparticles in a broad spectrum of different reactions such as inter alia aldol condensations and hydrogenation reactions For example in the aqueous phase hydrogenation of the biomass based key platform chemical levulinic acid into valerolactone and beyond employing heterogeneous catalysts and nanoparticles the presence of water has a beneficial effect and accelerates the reaction rates whereas in organic solvents much lower activities were observed This promotional effect of water in the hydrogenation of levulinic acid was proved by many experimental and theoretical studies using a broad spectrum of different types of catalytic systems The Software Encyclopedia, 1988 Index Medicus .2004 Vols for 1963 include as pt 2 of the Jan issue Medical subject headings LAB 2 Enzyme Catalysis Cathrine M. Monson, 2004*

Enzyme Catalysis and Regulation Gordon G. Hammes,1982 Enzyme Catalysis and Regulation Gorden Hammes,2012-12-02 Enzyme Catalysis and Regulation is an introduction to enzyme catalysis and regulation and covers topics ranging from protein structure and dynamics to steady state enzyme kinetics multienzyme complexes and membrane bound enzymes Case studies of selected enzyme mechanisms are also presented This book consists of 11 chapters and begins with a brief overview of enzyme structure followed by a discussion on methods of probing enzyme structure such as X ray crystallography and optical spectroscopy Kinetic methods are then described with emphasis on the general principles of steady state and transient kinetics The chemical principles involved in enzyme catalysis are also discussed and case studies of a few well documented enzymes are presented The regulation of enzyme activity is analyzed from a nongenetic viewpoint with particular reference to binding isotherms and models for allosterism Two particular enzymes aspartate

transcarbamoylase and phosphofructokinase are used as examples of well studied regulatory enzymes. The last two chapters focus on multienzyme complexes and membrane bound enzymes. This monograph is intended for graduate students advanced undergraduates and research workers in molecular biology and biochemistry.

Molecular Aspects of Enzyme Catalysis. Toshio Fukui, 1994

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web 01 identify the acid and base involved in the reaction 02 write down the chemical formula of the acid and base 03

determine the balanced equation by ensuring the number of atoms on both sides of the equation is equal 04 balance the charges by adding any necessary coefficients to balance the equation 05

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