

Properties of Solutions: Electrolytes and Non- Electrolytes

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Course: CHE121

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Abstract

The purpose of this lab experiment was discovering properties of strong electrolytes, weak electrolytes and non- electrolytes by using the conductivity of solutions to check for these properties, to determine which molecules or ions are responsible for the conductivity of the solutions and to write the equations for the dissociation of the compounds in water. Investigate the conductivity of solutions resulting from compounds that dissociate to produce different numbers of ions. This was accomplished by submerging a conductivity probe into test tubes containing 0.05M of the solutions, waiting for the values to stabilize on the computer monitor and recording the value. The conductivity of each solution was found to be CaCl_2 - 8460, AlCl_3 -11514, NaCl -5567 $\text{HC}_2\text{H}_3\text{O}_2$ -451, HCl -13746, H_3PO_4 -7321, H_3BO_3 -44, $\text{H}_2\text{O}_{\text{distilled}}$ -43, $\text{H}_2\text{O}_{\text{tap}}$ -351, CH_3OH -44 And the dissociation of each solution was;

$\text{CaCl}_2(\text{aq}) \rightleftharpoons \text{Ca}^{2+}(\text{aq}) + 2\text{Cl}^{-}(\text{aq})$	$\text{H}_3\text{PO}_4(\text{aq}) \rightleftharpoons \text{H}^{+}(\text{aq}) + \text{H}_2\text{PO}_4^{-}(\text{aq})$
$\text{AlCl}_3(\text{aq}) \rightleftharpoons \text{Al}^{3+}(\text{aq}) + 3\text{Cl}^{-}(\text{aq})$	$\text{H}_3\text{BO}_3(\text{aq}) \rightleftharpoons \text{H}^{+}(\text{aq}) + \text{H}_2\text{BO}_3^{-}(\text{aq})$
$\text{NaCl}(\text{aq}) \rightleftharpoons \text{Na}^{+}(\text{aq}) + \text{Cl}^{-}(\text{aq})$	$\text{H}_2\text{O}(\text{aq}) \rightleftharpoons \text{H}^{+}(\text{aq}) + \text{OH}^{-}(\text{aq})$ distilled water
$\text{HC}_2\text{H}_3\text{O}_2(\text{aq}) + \text{H}_2\text{O}(\text{aq}) \rightleftharpoons \text{C}_2\text{H}_3\text{O}_2^{-}(\text{aq}) + \text{H}_3\text{O}^{+}(\text{aq})$	$\text{H}_2\text{O}(\text{aq}) \rightleftharpoons \text{H}^{+}(\text{aq}) + \text{OH}^{-}(\text{aq})$ tap water

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web 1 name location of proposed ministry and leadership in place 2 using missioninsite information on your target area demonstrate the ministry potential for the area and include the following 2 1 what are the geographic boundaries of the target area 2 2 what is the overall population of this mission field