Lab: Baking Soda Stoichiometry

Introduction: When baking soda (NaHCO₃) is an ingredient in your recipe, its purpose is to make the batter rise and produce a product with a light and fluffy texture. The decomposition reaction is shown below:

Our goal will be to predict how much sodium carbonate (Na₂CO₄) is produced when baking soda decomposes.

Materials: Lab Video (use to fill in data table)

Procedure:

- 1. Use the lab video to complete the data table below.
- Calculate the mass of the NaHCO₃ used in the reaction and the experimental mass of Na₂CO₃ that was
 produced. You need to show work in the "calculation" table for how you got your answer. Typing it
 out is okay for the calculation table section.
- 3. Answer the 2 conclusion questions at the end.

Data table

Mass of crucible (g)	12.74 g
Mass of crucible and NaHCO ₃ (g)	14.33 g
Mass of crucible and Na ₂ CO ₅ (g)	13.75 g

Calculations Table (show work in each box)

Mass of NaHCO ₃ (g)	1.59 g	
Experimental (actual) mass of Na2CO3 (g)	1.01 g	

Conclusion Questions: Show all work, handwritten, using the correct steps/methods shown during class. You will need to upload an image of your work.

 Use your starting mass of NaHCO, and the balanced chemical equation above to calculate the mass of Na,CO, that should be produced by the decomposition reaction. This is the theoretical yield.

Baking Soda Stoichiometry Lab Answers

NARAYAN CHANGDER

Baking Soda Stoichiometry Lab Answers:

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The step by step instructions are easy to follow and safety precautions are clearly highlighted throughout the book Visual aids such as illustrations and photographs are included to enhance understanding and engagement Beyond the individual experiments we encourage a spirit of inquiry and exploration guiding children to ask questions form hypotheses and analyze their results developing essential scientific skills We believe that learning should be an exciting adventure and we hope this book will ignite a lifelong passion for science in you and your children Let's embark on this incredible journey of scientific discovery together Prepare to be amazed Working with Chemistry Donald J. Wink, Sharon Fetzer-Gislason, Julie Ellefson Kuehn, 2004-02-20 With this modular laboratory program students build skills using important chemical concepts and techniques to the point where they are able to design a solution to a scenario drawn from a professional environment The scenarios are drawn from the lives of people who work with chemistry every day ranging from field ecologists to chemical engineers and include many health professionals as well Teaching Science for Understanding James Joseph Gallagher, 2007 Offers middle and high school science teachers practical advice on how they can teach their students key concepts while building their understanding of the subject through various levels of learning activities Improving Student Comprehension of Stoichiometric Concepts Connie Lynn Bannick Kemner, 2007 The Science Teacher, 2005 SCC Library Molecular Level Drawings as a Road Map for Conclusion Writing in a High School Classroom Patrick has 1964 cur Mackey Bohman, 2008 Holt Chemistry Ralph Thomas Myers, 2004 Subject Guide to Books in Print, 1997 Α **Stoichiometry Unit** David Callaghan, 2004 Stoichiometry Unit Project Luann Marie Decker, 1998

Baking Soda Stoichiometry Lab Answers Book Review: Unveiling the Power of Words

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