Carr's smaller considerate Act 11

Introduction:

When we bake/cook semething, we use a specific amount of each ingredient. Imagine if you made a batch of cookies and used was too many eggs, or not amough sugar. YUCK! In alternistry, reactions proceed with very specific needers. The study of those scorpes is accordance by. When the reactions are precent in the cornect amounts, the species mades products. We are point to look at what happens when there are more or less of some of the reactions present.

Propodure: FACT Mandatana & Pilly total the Stead & Chambers & Associate, Produce, and Expressed ONES INSTRUCTS

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Part 1: Making Sandwiches:



- A chouse sandwick requires 2 slices of broad to 1 slice of chaese. If there is exactly retire as many slices of broad as choose, then there will be no letterers. What happens if you have two slices of broad and two slices of chaese? You have a slice of choose left event.
- Use the signalation to play proposed with this two-voyescent symbols a neartism. Take some time and familiarities required (with the signalation business).
- At the bottom of the screen, change the coefficients of the tructums to 2.1.



- 4. Notice that there is now one sundwich with no leftowers!
- 5. Complete situations il through D. A has been done for you as an avairable.

Situation	Bread Used	Cheese li sed	Sandwiches Made	Excess Broad	- Excess Cheese
A.	it choes	5 stines	2 sandwiches	O bread	3 oboves
D-	2 sticus	2 stiges			
- C			2 stadwiches	Littles	H Africa
100	- 6 altern		3 sandwiches	11.000	4 shops

- Look at situation A. The choose in this case is said to be in excess. The other reaction (bread) has no leftovers. The bread in this case is the Knobbey reactions.
- 7. In situations B. C. and D. determine the Emitting resolute: 18: ______ C: _____ D: _____

Part 2: Real Chemical Reactions:



- Scient "Molecules" towards the bottom of your screen. We can now work with real characteristics one that entries a very entertaining BOOM?
- 9 What is the lowest molecule ratio for the reaction of by drogen and croygen to produce water? That is, halance the following equation:

$$\underline{M}_0 + \underline{O}_0 \rightarrow \underline{M}_1 O$$

10. Complete the table believ white making water HJO from hydrogen H₂ and one per O₂

Hydrogen Molecules H ₁	Oscigen Malestales Oc	Water Molecules 85/0	Excess H ₀	Excess O ₂
4 molecules	4 molecules			
7 molecules	6 molecules			
		4 molecules	0 molecules	If medication
6 molecules	8 molecules			
		4 molecules	I melocales	 If the location

Stag for check point. Teacher signature for this page:

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