Inorganic Photochemistry – III

Photochromism

It is the reversible transformation of a compound between two forms by absorption of electromagnetic radiation, where the two forms have different absorption spectra. Simply it can be described as a reversible change in color on exposure to light. The phenomenon was first observed in 2,3,4,4-tetrachloro naphthalene-1-one in the solid state.

In many cases there is an absorption band in only one form. The degree of change required for a photochemical reaction to be called photochromic is a dramatic change in color. Thus while the cis-trans isomerization of azobenzene is photo chromic that stilbene is not

Trans stilbene

eis stilbene

Since photochromism is just a special case of photochemical reaction almost any photochemical reaction like cis-trans isomerization, pericyclic reaction, intramolecular H transfer, electron transfer etc can be photochromic

A requirement of photochromism is that the two states must be thermally stable. All photochromic molecules isomerize back to their more stable form at some rate and is accelerated by heating.

Many inorganic compounds show photochromism with better fatigue resistance (i.e loss of sensitivity due to photo oxidation, degradation etc eg AgCL is used in making photochromic lenses. Other silver and Zinc halides are also photochromic.

Photochromic complexes are rare. Two classes are sodium nitroprusside and ruthenium sulfoxide. The mode of action is an excited isomerization of sulfoxide ligand on Ru polypyridine from S to O or O to S leads to a color change and a change in Ru III/II. The ground state is always S bonded and metastable state O bonded.

Inorganic Photochemistry Lecture Notes

Padhraic Smyth

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Pure and Applied Chemistry, 1970 Some vols include Plenary lectures of the International Conference on Organometallic Chemistry **Emerging Nanomaterials for Energy Conversion and Storage Applications** Guohua Choice ,2008 Jia, Hongbo Li, Zongyou Yin, 2022-01-04 The Cumulative Book Index ,1999 Molecular Distortions in <u>Ionic and Excited States</u> Peter V. Schastnev, Lyudmila N. Shchegoleva, 1995-08-25 This book discusses the achievements in the study on the structure of active species being formed at the change of molecule charge or electronic state It gives a systematic outline of the problem of molecular structure distortions in radical ionic and excited states involving experimental and theoretical material The text focuses on analyzing the physical reasons for structural distortion occurrences at a model level and also with results of detailed quantum chemical calculations. The book presents numerous facts on the structural distortions in ions of various types of organic molecules Data is presented for the first time on distortions in radical ions of polyfluoroaromatics and other conjugated molecules **Cumulated Index to the Books** ,1999 American Book Publishing Record, 2001 Nuclear Science Abstracts ,1974 The Stratosphere and Mesosphere: Principal lectures and seminars .1976 Concepts of Inorganic Photochemistry Arthur W. Adamson, Paul D. Fleischauer, 1975 PHOTOPHYSICAL PROCESSES ENERGY LEVELS AND SPECTRA KINETICS OF PHOTOPHYSICAL PROCESSES CHARGE TRANSFER PHOTOCHEMISTRY SUBSTITUTIONAL PHOTOCHEMISTRY OF FIRST ROW TRANSITION ELEMENTS PHOCHEMISTRY OF THE HEAVIER ELEMENTS PHOTOCHEMISTRY OF CARBONYL COMPLEXES PHOTOCHEMISTRY OF 1 3 DIKETONATE CHELATES THE PHOTOLYSIS OF SIMPLE INORGANIC IONS IN SOLUTION PHOTOCHEMISTRY IN THE SOLID STATE PHOTOCHROMISM AND CHEMILUMINESCENCE Russian Chemical Reviews ,2007 *Inorganic* Photochemistry Rudi van Eldik, Grazyna Stochel, 2011-07-27 The Advances in Inorganic Chemistry series present timely and informative summaries of the current progress in a variety of subject areas within inorganic chemistry ranging from bio inorganic to solid state studies This acclaimed serial features reviews written by experts in the field and serves as an indispensable reference to advanced researchers Each volume contains an index and each chapter is fully referenced

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