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web introduction molecular shape and polarity greatly influence the chemical and physical properties of compounds because a molecule will always assume the geometry that gives it the lowest potential energy molecular shape can

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web draw lewis structure use vsepr to determine molecular geometry determine bond polarity based on electronegativity differences determine molecular polarity based on bond dipoles molecular geometry for the following molecules complete this

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web two regions of electron density around a central atom in a molecule form a linear geometry three regions form a trigonal planar geometry four regions form a tetrahedral geometry five regions form a trigonal bipyramidal geometry and six regions form an octahedral geometry

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web molecules composed of covalently bonded atoms may also be polar or nonpolar for the molecule to be polar it must of course have polar bonds but the key factor for determining the polarity of a molecule is its shape if the polar bonds dipoles are symmetrical around the central atom they offset each other and the resulting molecule is

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web linear geometry all angles 180 degrees three regions of electron density around a central atom in a molecule form a trigonal planar geometry all angles 120 degrees four regions of electron density around a central atom in a molecule form a tetrahedral geometry all angles 109.5 degrees

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web in this investigation you will examine a theory that chemists use to explain different aspects of chemical bonding valence shell electron pair repulsion vsepr theory attention will be given to how molecules are arranged in different shapes and how chemists can predict the geometry of a given molecule