



## Lab Data



**Verify** your calculation. In your set-up, do your units cancel?

Did you report your data to the correct number of significant figures?

Mass of magnesium (g)	0.015
Moles of magnesium (mol)	0.00062
Temperature of water (°C)	24.0
Temperature of water (K)	297.15
Vapor pressure of water (mmHg)	22.4
Barometric pressure (mmHg)	738.0

Observations

bubbles

Volume of hydrogen gas collected (mL)	15.00
Volume of hydrogen gas collected (L)	0.00049
Pressure of hydrogen gas (mmHg)	750.2

LAB DATA

# Introduction To Gas Law Lab Answer Key

**David J. WEINER**



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