

SPORTS LEAGUE BUDGET

Sports League Name				Income			
Tournament/League Season				Expenses			
2024				Difference			
EXPENSES				INCOME			
Rental + Registration Fees				Per Team Fees			
Budget Item	Budgeted Amount	Actual Amount	Difference	Item	Total Amount	# of Teams	Per T
Reservation	\$ 500.00	\$ 400.00	\$ 100.00	Field Usage	\$ 1,200.00	+	\$
	\$ 1,000.00	\$ 1,200.00	\$ (200.00)	Referee	\$ 1,000.00	+	\$
Red Cards	\$ 400.00	\$ 500.00	\$ (100.00)	Team ID per player	\$ 500.00	+	\$
	\$ 500.00	\$ 400.00	\$ (100.00)	Yellow + Red Card	\$ 25.00	+	\$
	\$ 800.00	\$ 700.00	\$ 100.00	Charitable donations	\$ 200.00	+	\$
Win- 1st Place	\$ 500.00	\$ 500.00	\$ -	Registration Fee	\$ 250.00	+	\$
Win- 2nd Place	\$ 250.00	\$ 250.00	\$ -	Trophies + Medals	\$ 600.00	+	\$
			\$ -	Other			
			\$ -	Other			
			\$ -	Other			
			\$ -	Other			
			\$ -	Other			
			\$ -	Other			
			\$ -	Other			
			\$ -	Other			
			\$ -	Other			
			\$ -	Other			
\$	3,950.00	\$ 4,150.00	\$ (200.00)	Total Per Team			\$



EXCEL SPREADSHEET | GOOGLE SHEETS



Sample Of A Budget For A Basketball Tournament

Thomas H. Sawyer



Sample Of A Budget For A Basketball Tournament:

The Process of Recreation Programming Patricia Farrell, Herberta M. Lundegren, 1983 Facility Design and Management for Health, Fitness, Physical Activity, Recreation, and Sports Facility Development Thomas H. Sawyer, 2005 Since 1946 the Council on Recreation and Physical Activity Venue Design and Management have endorsed Facility Design and Management This 11th edition adds a number of new chapters including financial management promotions public relations merchandising and ticket operations programming and event and risk management The book will have the following online resources Instructor s Guide and PowerPoint Presentation 34 Additional Appendices Author Biographies Glossary of Terms VersaChad Software Design Program with Custom Exercises *Juvenile Diversion* Kevin E. O'Brien, 1977 *Win at Any Cost* Francis X. Dealy, 1990 Facility Management for Physical Activity and Sport Thomas H. Sawyer, 2009 In an era of unprecedented expansion sport has become even more significant and pervasive in our society Through this growth a new breed of specialist has emerged the public assembly facility manager Facility Management for Physical Activity Sport provides managers with a fundamental understanding of the concepts and issues facing today s professional *The Money Game* Robert H. Atwell, Bruce Grimes, Donna A. Lopiano, 1980 *Organization for Championship High School Basketball* Sal Verderame, 1963 **The Futurist** , 2001 **Ebony** , 1962-12 EBONY is the flagship magazine of Johnson Publishing Founded in 1945 by John H Johnson it still maintains the highest global circulation of any African American focused magazine Accounting Procedures for Student Organizations California. State Department of Education. School District Management Services Bureau, 1988 **Proceedings of the ... Annual Convention** National Collegiate Athletic Association. Convention, 1960 *The IMC planPro handbook* Kenneth E. Clow, Donald Baack, 2007 This supplemental package consists of booklet and disk and is available from Prentice Hall The IMC Plan Pro disk provides an exercise for a student or business person to prepare an entire marketing communications program The booklet provides step by step instructions about how to use the disk and gives brief explanations of the IMC concepts that are part of the program

Financing the Sport Enterprise Thomas H. Sawyer, Michael G. Hypes, Julia Ann Hypes, 2004 Designed to meet the standards established by the National Association for Sport and Physical Education NASPE and the North American Society for Sport Management NASSM this book is intended to provide professional practitioners and students in sport management a book with comprehensive survey coverage of the many traditional and innovative aspects of financial management The focus of this text is upon the basics of financial management including but not limited to diversification of revenue acquisition of revenue resources financial management auditing sponsorships licensing franchises fund raising and more Different from most general finance texts this book focuses on and uses many examples specific to the sports field Plus it has many practical examples and ideas for the practitioner to use With this knowledge readers will be able to take a proactive role in confronting the fiscal challenges faced by their organisation *Association Men* , 1924 Colorado State University. Department of

Intercollegiate Athletics CliftonLarsonAllen LLP.,2019 *The Successful High School Athletic Program* Robert L. Purdy,1973 Administration of School Health and Physical Education Programs Charles Augustus Bucher,1963
Administration of Health and Physical Education Programs, Including Athletics Charles Augustus Bucher,1975
Athletic Journal ,1927 Vols 9 10 include proceedings of the 8th 11th annual meeting of the American Football Coaches Association and of the 3d 6th annual meeting of the National Association of the Basketball Coaches of the United States
Middle-west School Review ... ,1919

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web here s some of the guidelines for drawing dot structures so let s say we wanted to draw the dot structure for this molecule so silicon tetrafluoride the first thing we would need to do is to find the total number of valence electrons and we would account for these valence electrons in our dot structure

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web the arrangement of atoms in several biologically important molecules is given here complete the lewis structures of these molecules by adding multiple bonds and lone pairs do not add any more atoms a the amino acid serine b urea c pyruvic acid d uracil e carbonic acid answer a answer b answer c answer d answer e

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web sep 12 2023 lewis dot structure is mainly a graphic representation of the valence shell electrons of different atoms in chemistry it was invented by a renowned physical chemist of america gilbert newton lewis 1875 1946 who had introduced and mentioned this in his 1916 published article under the topic the atom and the molecule

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web a lewis diagram shows how the valence electrons are distributed around the atoms in a molecule shared pairs of electrons are drawn as lines between atoms while lone pairs of electrons are drawn as dots next to atoms

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web lewis diagrams ap chem sap 4 eu sap 4 a lo sap 4 a 1 ek google classroom you might need periodic table ethanethiol $\text{C}_2\text{H}_6\text{S}$ is a clear liquid with a strong odor the compound is often added to otherwise odorless fuels such as natural gas to help warn of gas leaks

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web answers lewis dot lewis dot structures for each molecule on the worksheet the lewis dot structure the number of valence electrons the electron arrangement e a and the molecular geometry m g are given respectively to see a larger view of the lewis dot structure point at the molecule right click on the mouse and go to view image

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web you should try to answer the questions without referring to your textbook if you get stuck try asking another group for help for each of the following draw the lewis dot structure give the electron arrangement e a and the molecular geometry m g

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web in 1916 american chemist gilbert n lewis introduced bond lines to electron dot structures these structures also known as lewis structures or electron dot structures are drawings that visually demonstrate how electrons are shared and arranged around atoms the electrons denoted as dots are called lone pairs and belong to an individual

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web lewis dot structures are useful for describing chemical bonds but have some flaws a lewis structure is a picture of a molecule that shows the covalent bonds and pairs of free electrons the octet rule is the basis for lewis structures

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web lewis dot formula also called an electron dot formula shows the valence electrons indicating the bonding between atoms the following guidelines will help draw the electron dot formulas correctly

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web lewis dot structures also called electron dot structures are diagrams that describe the chemical bonding between atoms in a molecule they also display the total number of lone pairs present in each of the atoms that constitute the molecule

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ions with one central atom introduction to lewis structures a lewis structure is a way to show how atoms share electrons when they form a molecule lewis structures show all of the valence electrons in an atom or molecule

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web lewis structures we also use lewis symbols to indicate the formation of covalent bonds which are shown in lewis structures drawings that describe the bonding in molecules and polyatomic ions for example when two chlorine atoms form a chlorine molecule they share one pair of electrons

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web this online quiz is intended to give you extra practice in identifying and drawing lewis dot structures as well as predicting ion formation this quiz aligns with the following ngss standard s hs ps1 1

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web jan 30 2023 a lewis structure is a very simplified representation of the valence shell electrons in a molecule it is used to show how the electrons are arranged around individual atoms in a molecule electrons are shown as dots or for bonding electrons as a line between the two atoms

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web jan 30 2023 follow these simple steps to draw lewis dot structures draw the atoms on paper and put dots around them to represent valence electrons of the atom be sure to have the correct number of electrons if the species is an ion add or subtract electrons corresponding to the charge of the ion

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web get the gizmo ready on the periodic table tab select ar argon select the electron configuration tab turn on show number of electrons question how are the electron configurations of elements beyond argon determined arrange create the correct electron configuration for argon

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web electron configuration answer key vocabulary atomic number atomic radius aufbau principle chemical family diagonal

rule electron configuration hund s rule orbital pauli exclusion principle period shell spin subshell prior knowledge questions
do these before using the gizmo

web the electron configurations worksheet 1 state the four quantum numbers and the possible values they may have
principal quantum number n 1 2 3 angular momentum quantum number l 0 to $n - 1$ magnetic quantum number m_l $-l$ to $+l$ spin
quantum number m_s $+\frac{1}{2}$ and $-\frac{1}{2}$

web answer there are 2 electrons in the k shell 8 electrons in the l shell 18 electrons in the m shell and 32 electrons in the n shell q10 which is the orbital occupied by the last electron in the boron element

web electron configuration atomic radius pm aluminum 13 1s 2 2s 2 2p 6 3s 2 3p 1 118 silicon 14 1s 2 2s 2 2p 6 3s 2 3p 2 111 phosphorus 15 1s 2 2s 2 2p 6 3s 2 3p 3 98 sulfur 16 1s 2 2s 2 2p 6 3s 2 3p 4 88 chlorine 17 1s 2 2s 2 2p 6 3s 2 3p 5 79 argon 18 1s 2 2s 2 2p 6 3s 2 3p 6 71

web photoelectron spectroscopy electron configurations google classroom you might need periodic table using s p d f notation what is the electron configuration for a neutral atom of beryllium choose 1 answer 1 s 2 2 s 2 a 1 s 2 2 s 2 1 s 2 2 s 1 2 p 1 b 1 s 2 2 s 1 2 p 1 1 s 2 2 p 2 c 1 s 2 2 p 2 1 s 2 1 p 2 d 1 s 2 1 p 2 show periodic table stuck

web ns f4 electron configuration answers ground state orbital diagrams and electron configurations compare the ground state orbital diagrams below to the boarding house diagrams on page 1 compare the electron configurations below to the manger s code also on page 1 6 examine the orbital diagrams and electron configurations as shown

web electron configuration guided notes answer key oxygen has an atomic number of 8 give its electron configuration answer an atomic number of 8 means a total of eight electrons two of these electrons can fit into the first shell while the remaining six can fit into the second shell therefore the configuration is written as 2 6

web the four different types of orbitals s p d and f have different shapes and one orbital can hold a maximum of two electrons the p d and f orbitals have different sublevels thus can hold more electrons as stated the electron configuration of each element is unique to its position on the periodic table

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web explanation the first two electrons fit into the 1s orbital 1s 2 then next two into the 2s orbital 2s 2 the 2p orbital has three slots which can take 2 electrons each a total of 6 which gives 2p 6 the final two electrons fit in the 3s orbital 3s 2
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web using the periodic table to determine the electron configurations of atoms is key but also keep in mind that there are certain rules to follow when assigning electrons to different orbitals the periodic table is an incredibly helpful tool in writing electron configurations

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web explain your answer the first electron to leave the atom would be one in the highest energy level which is the fourth energy level electrons in the highest energy level are the least attracted to the nucleus because they are the most distant 25
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web electron configurations worksheet with answers electron configuration notation shows the arrangement of electrons around the nucleus of an atom helps chemist understanding how elements form chemical bonds can be written using the periodic table or an electron configuration chart

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web the following electron configurations belong to which elements 21 1s22s22p63s1 sodium 22

1s22s22p63s23p64s23d104p65s24d6 ruthenium 23 kr 5s24d10 cadmium 24 xe 6s24f145d106p2 lead 25 rn 7s25f146d4 seaborgium determine if the following electron configurations are correct 26 1s22s22p63s23p64s24d104p65s1 no it should be 3d10

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web how many protons neutrons and electrons are in atoms of these isotopes write the complete electron configuration for

each isotope answer co has 27 protons 27 electrons and 33 neutrons 1s 2 2s 2 2p 6 3s 2 3p 6 4s 2 3d 7 i has 53 protons 53 electrons and 78 neutrons 1s 2 2s 2 2p 6 3s 2 3p 6 3d 10 4s 2 4p 6 4d 10 5s 2 5p 5

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web select the electron configuration tab on show number of electrons introduction beyond argon it is a bit tricky to determine which subshell gets filled next there are several rules that scientists use to determine the electron configurations of larger atoms question how are the electron configurations of elements beyond argon determined

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