

Stoichiometry Review

Directions: Each of the problems below reviews a different type of stoichiometry. Be sure to read the problem carefully to know what it is asking for.

Mass-to-Mass Problems

1. How many grams of potassium chloride, KCl, are produced during the decomposition of 100.0 grams of potassium chlorate, KClO₃?



$$\frac{100.0 \text{ g KClO}_3}{1} \times \frac{1 \text{ mol KClO}_3}{122.545 \text{ g KClO}_3} \times \frac{1 \text{ mol KCl}}{1 \text{ mol KClO}_3} \times \frac{74.548 \text{ g KCl}}{1 \text{ mol KCl}} = \underline{60.83 \text{ g KCl}}$$

2. Find the number of grams of HCl needed to react completely with 0.50 moles of magnesium.



$$\frac{0.5 \text{ mol Mg}}{1} \times \frac{1 \text{ mol HCl}}{1 \text{ mol Mg}} \times \frac{36.458 \text{ g HCl}}{1 \text{ mol HCl}} = \underline{18.23 \text{ g HCl}}$$

3. Find the mass in grams of hydrogen gas produced when 14.0 moles of HCl are added to an excess amount of magnesium.



$$\frac{14 \text{ mol HCl}}{1} \times \frac{1 \text{ mol H}_2}{2 \text{ mol HCl}} \times \frac{2.016 \text{ g H}_2}{1 \text{ mol H}_2} = \underline{14.112 \text{ g H}_2}$$

4. If 0.75 moles of ammonia are needed, how many grams of nitrogen will be consumed?



$$\frac{0.75 \text{ mol NH}_3}{1} \times \frac{1 \text{ mol N}_2}{2 \text{ mol NH}_3} \times \frac{28.014 \text{ g N}_2}{1 \text{ mol N}_2} = \underline{10.51 \text{ g N}_2}$$

Review Stoichiometry Section 1 And 2 Answers

Luann Marie Decker



Review Stoichiometry Section 1 And 2 Answers:

A Complete Preparation for the MCAT Aftab S. Hassan, James L. Flowers, 1992 This guide for MCAT preparation applies the principles of active and problem based learning to an updated review of content and skills with models for enhanced problem solving and critical thinking abilities There are details on setting up a self managed study programme with guidelines for time management and stress management All areas tested on the exam are covered verbal reasoning physical science writing sample biological sciences with practice questions to chart progress **Complete Preparation for the MCAT** Williams & Wilkins Review, 1998-04 Here is the most respected test prep book for the Medical College Admission Test you can buy featuring an active learning approach for a better understanding of the exam s content and a better chance for success Unique to this guide are coverage of all recent changes in the MCAT plus a step by step plan for sharpening cognitive skills developing problem solving skills and critical thinking This thorough guide replaces expensive test preparation courses while giving students exactly what they need to get ready for the MCAT *Open-Ended Problems* James Patrick Abulencia, Louis Theodore, 2015-03-23 This is a unique book with nearly 1000 problems and 50 case studies on open ended problems in every key topic in chemical engineering that helps to better prepare chemical engineers for the future The term open ended problem basically describes an approach to the solution of a problem and or situation for which there is not a unique solution The Introduction to the general subject of open ended problems is followed by 22 chapters each of which addresses a traditional chemical engineering or chemical engineering related topic Each of these chapters contain a brief overview of the subject matter of concern e g thermodynamics which is followed by sample open ended problems that have been solved by the authors employing one of the many possible approaches to the solutions This is then followed by approximately 40 45 open ended problems with no solutions although many of the authors solutions are available for those who adopt the book for classroom or training purposes A reference section is included with the chapter s contents Term projects comprised of 12 additional chapter topics complement the presentation This book provides academic industrial and research personnel with the material that covers the principles and applications of open ended chemical engineering problems in a thorough and clear manner Upon completion of the text the reader should have acquired not only a working knowledge of the principles of chemical engineering but also and more importantly experience in solving open ended problems What many educators have learned is that the applications and implications of open ended problems are not only changing professions but also are moving so fast that many have not yet grasped their tremendous impact The book drives home that the open ended approach will revolutionize the way chemical engineers will need to operate in the future

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Holt Chemistry Ralph Thomas Myers, 2004

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throughout the text provides students with a more modern approach to analytical chemistry *Basic Concepts of Chemistry, Study Guide* Leo J. Malone, 2000 The text's three main goals are to introduce chemistry as a living relevant science to encourage learning and critical thinking and to help readers overcome the math difficulties that impede their progress in chemistry Designed to help readers master the principles of general chemistry As a prep book it promotes active involvement with the material There are special features throughout that reinforce concepts and help to develop strong problem solving and study skills Updated to Include an Interactive Learning Ware problems CD containing several of the chapter ending problems from the book in an interactive tutorial with feedback to help readers set up and solve problems

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