

# Chapter 5

# Periodic Table

# Periodic Table of the Elements

The periodic table is organized into groups and periods. The elements are color-coded as follows:

- Group 1 (Alkali Metals):** Orange
- Group 2 (Alkaline Earth Metals):** Yellow
- Groups 3-10 (Transition Metals):** Various shades of blue and purple
- Group 11 (Coinage Metals):** Green
- Group 12 (Zinc Group):** Light Green
- Groups 13-18 (Main Group Elements):** Various shades of green and yellow
- Lanthanides and Actinides:** Various shades of blue and purple

The periodic table is a tabular arrangement of the chemical elements, organized on the basis of their atomic numbers, electron configurations, and recurring chemical properties. The table is divided into sections for Lanthanides and Actinides at the bottom.

## The Periodic Table

- A map of the building block of matter.

The image shows a periodic table of elements. A callout box for Hydrogen (H) highlights the following information:

- Atomic number:** 1
- Symbol:** H
- Atomic weight:** 1.01 (rounded value)

The periodic table includes the following groups and elements:

- Group IIA:** Be, Mg, Ca, Sr, Ba, Ra
- Group IIIA:** B, Al, Ga, In, Tl
- Group IVA:** C, Si, Ge, Sn, Pb
- Group VA:** N, P, As, Sb, Bi
- Group VIA:** O, S, Se, Te, Po
- Group VIIA:** F, Cl, Br, I, At
- Group VIIIA:** He, Ne, Ar, Kr, Xe, Rn

The periodic table also includes the following elements:

- Hydrogen (H):** 1.01
- Lithium (Li):** 6.94
- Sodium (Na):** 22.99
- Potassium (K):** 39.10
- Rubidium (Rb):** 85.47
- Cesium (Cs):** 132.91
- Francium (Fr):** 223.02
- Helium (He):** 4.00
- Neon (Ne):** 20.18
- Argon (Ar):** 39.95
- Krypton (Kr):** 83.80
- Xenon (Xe):** 131.29
- Radon (Rn):** 222.02

# Chapter 5 The Periodic Table Assessment Answers

**Peter P. Berlow**



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