Chapter 6

# **CHEMICAL BONDING**

### MULTIPLE CHOICE QUESTIONS

	(a)	HCI	(b)	ecules has a co	(c)	NH <sub>4</sub> CI	TOTAL LANGE	(d)	AlkCla	(LHR 05)
2. 3.		ngle formed i			4.00			400		(GRW 06)
	(a)	120°	(b)	180°	(c)	109.50		(df)	107.5	
	Ionic compounds are mostly obtained by the combination of group:							(GRW 97)		
	(a)	3 and 5	(b)	2 and 5	(c)	4 and 3		(d)	1 and 2	
4.		e moment of		2.0000	10,000	75,365,564		100	-	(GRW 97)
	(a)	1.84D	(b)	Zero D	(c)	0.95 D		(d)	2.2 D	
5.				bitals are orien		Control and their Basis		1	ACTUAL NAME	(LHR 07)
	(a)	109.50	(b)	120°	(c)	1800		(d)	00	(Carried Co.)
6.	Which of the following species has unpaired electron in anti bonding molecular orbitals?									
7.	(MTN 08, 10, BWP, LHR 08									
	(a)	Ho	(b)	Hes	(c)	Os = 2		(d)	Maria	
				tain maximum			all trees	(m)		(LHR 08)
Acres 1	(a)	One Online	(b)	Two	(c)	Three	1000	(dl)	Four	( Commence of the last
8.	The state of the s	The state of the s	No. 20 (19)	ne have dipole i	ACCOUNTS OF THE PARTY OF	The second secon		1,007	1 Section	(GRW 08)
2000	(a)	Zero and 1.8		1.70 D and 1.8		(c)		ave zero	(d)	None of these
9.				n stability and					(0)	CHOWARD I
	(a)	They are ven		in stability and	HOUSE F					
						(b)				complete
	(c) They are gases (d) They are present in zero group Which species has unpaired electron in anti-bonding molecular orbits?									
10.							tecutair		0.02000	(GRW 09)
	(a)	O2+2	(b)	N <sub>2</sub> 2-	(c)	Ba		(d)	Fa	
11.				n the formation						11, GRW 09, 12)
	(a)	NE <sub>3</sub>	(b)	CF4	(c)	CCI		(d)	PCI <sub>5</sub>	
12.	The n	rost stable el-		hre:						(LHR 09)
	(a)	Halogens	(b)	Lithium family	(c)	Noble :	gases	(d)	None o	
13.	The h	ybridization o	of carbon	in C2H4 is:						<b>GRW 09, LHR 14</b>
	(a)	SD	(b)	SD <sup>3</sup>	(c)	sp <sup>3</sup>		(d)	mot hyd	widized
14.	Positi	ve ions are ca	illed:			1000		797		(LHR 09)
	(a)	Cations	(b)	Anions	(c)	Molecu	des	(d)	Hydrate	ed ions
15. 16.		number of bo	ends in C	H <sub>4</sub> molecule is				ALC: NO		TURIS FOR
	(a)	Six	(b)	Four	(c)	Friend		(d)	Elight	
		.I unit of dipo		The state of the s	4.47					(LHR 10, 11)
17.	(a)	loude	(b)	Debye	(c)	Conton	ob roote	r (dl)	Nimo <sup>-2</sup>	
	- 「後不能のは、大き元元元氏」というというとは、後不能ののは、元十元を表示している。して、1、1、1、1、1、1、1、1、1、1、1、1、1、1、1、1、1、1、1									(GRW 11)
18.	(a)	SO <sub>2</sub>	(b)	NH <sub>4</sub> CI	(c)	CaHa		(d)	H <sub>2</sub> O	(12121-112)
				e element in th			on Toron	(w)	FREEZY.	(FSD, GRW 11)
19.						Chlorin		Assets in	Buorine	
	(a)	Oxygen	(b)	Nitrogen	(c)	Cheoren	HC:	(d)	Hanoulus	
DF -		angle between			4.4	and the second		45-465	Parties.	(LHR 10)
23.5	(a)	105.5°	(b)	107.5°	(c)	9/2°		(d)	95°	
20.		nit of dipole n			900	-			7227	(LHR 10)
	(a)	pm	(b)	Debye	(c)	m/C		(dl)	All	
21.	The amount of energy released by absorbing electron in the valence shell is:									
	(a)	Ionization en		Electron affinity	y (¢)	Electro	-negativ	ity (d)	Atomiza	ation energy
22.	The n	umber of elec	trons si	named in SFe:				89334		(GRW 10
	(a)	4	(b)	12	(c)	6		(df)	8:	

## **Chemical Bonding Section 1 Quiz Answers**

**Cecie Starr, Ralph Taggart** 

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