

# CHEMISTRY 1

## FINAL EXAM REVIEW

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Name \_\_\_\_\_ Section \_\_\_\_\_ Date \_\_\_\_\_

### Chemistry Review: Writing and Balancing Equations

**Directions:** Solve each problem and answer each question

1. Balance each equation below by adding the correct coefficients

- $\_\_\_ \text{CuCl}_2 + \_\_\_ \text{NaBr} \rightarrow \_\_\_ \text{NaCl} + \_\_\_ \text{CuBr}_2$
- $\_\_\_ \text{HCN} + \_\_\_ \text{Mg(OH)}_2 \rightarrow \_\_\_ \text{Mg(CN)}_2 + \_\_\_ \text{H}_2\text{O}$
- $\_\_\_ \text{HNO}_3 + \_\_\_ (\text{NH}_4)_2\text{SO}_4 \rightarrow \_\_\_ \text{NH}_4\text{NO}_3 + \_\_\_ \text{H}_2\text{SO}_4$
- $\_\_\_ \text{ZnCO}_3 + \_\_\_ \text{Al}_2\text{O}_3 \rightarrow \_\_\_ \text{ZnO} + \_\_\_ \text{Al}_2\text{CO}_3$
- $\_\_\_ \text{C}_4\text{H}_{10} + \_\_\_ \text{O}_2 \rightarrow \_\_\_ \text{CO}_2 + \_\_\_ \text{H}_2\text{O}$

2. Write and balance each equation. Include phase labels

- Aqueous sodium hydroxide reacts with aqueous sulfuric acid to form aqueous sodium sulfate and water
- Solid iron (III) hydroxide and aqueous sodium carbonate are formed from the reaction of solid iron (III) carbonate and aqueous sodium hydroxide
- Solid aluminum reacts with aqueous hydrobromic acid to form aqueous aluminum bromide and hydrogen gas
- Gaseous ethane (dibarbon hexahydride) reacts with oxygen gas to form gaseous carbon dioxide and gaseous water
- Solid calcium phosphite reacts with aqueous copper (II) nitrate to form aqueous calcium nitrate and solid copper (II) phosphite

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# 21 CHEMISTRY WORKSHEETS

## WORKED ANSWER KEY

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