

Review for Ch. 8 Test

8.1 Describing Chemical Reactions

- know how to write word equations (i.e. you have be able to name compounds, including acids, ionic, and covalent compounds from Ch. 7)
- know how to write a formula equation
- What 2 things do coefficients represent in an equation? **Relative number of moles of a substance ; relative number of molecules needed for a reaction to occur**
- know common symbols used in a reaction
- know 4 evidences of a chemical reaction:
 1. **energy in the form of heat or light**
 2. **production of a gas**
 3. **change in color**
 4. **formation of a precipitate**
- know the Law of conservation of matter: **matter can neither be created nor destroyed in a chemical reaction (which means, mass of reactants must equal the mass of products)**
- know what a chemical equation represents: **relative amounts of reactants and products**
- know how to balance chemical equations (combustion reactions are particularly tricky, practice!)

1. Complete the equation and write a word equation for the following reaction: $\text{ZnO} + \text{FeSO}_4 \rightleftharpoons$



Zinc oxide reacts with iron (II) sulfate to yield zinc sulfate and iron (II) oxide

2. Write a formula equation for the reaction of a solution of AgCl with potassium metal:



3. What elements are always diatomic gases at room temperature?

All of the halogens (Group 17) , H_2 , N_2 , O_2

4. Can you balance these reactions:



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Chemistry Success in 20 Minutes a Day Michael B. McGinnis, 2005 Offers a diagnostic test and twenty lessons covering vital chemistry skills **Earth's Surface: Teacher's ed**, 2005 Books in Print Supplement, 2002

Cumulative Book Index, 1925 A world list of books in the English language *The United States Catalog*, 1925

4th NASA/DoD Conference on Evolvable Hardware (EH 2002) Adrian Stoica, 2002 The purpose of this conference is to bring together leading researchers from the evolvable hardware community representatives of the automated design and programmable reconfigurable hardware communities technology developers and end users from the aerospace military and commercial sectors The focus of EH 2002 is evolvable hardware for safer systems

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