CHAPTER 9 REVIEW

Stoichiometry

Teacher Notes and Answers

Chapter 9 SECTION 1

SHORT ANSWER

- 1. 60
- 2. d.
- 3. a.
- 4. 6
- 5. a. $2N_2O(g) + 3O_2(g) \rightarrow 4NO_2(g)$
 - b. 4 mol NO₃:3 mol O₂
 - e. 15.0 mol
 - d. True
 - e. False
- a. 28.0 g/mol N₂
 2.0 g/mol H₂
 17.0 g/mol NH₃
 - b. 3 mol H₃:1 mol N₂; 2 mol NH₃:1 mol N₂; 2 mol NH₃:3 mol H₃; or their reciprocals
- 7. a. 1 mol NO:1 mol H:O
 - b. 3 mol NO:2 mol NH₃
 - c. 0.360 mol
- a. 4 mol O₂:1 mol C₃H₄; 3 mol CO₂:1 mol C₃H₄; 2 mol H₂O:1 mol C₃H₄;
 3 mol CO₂:4 mol O₂; 2 mol H₂O:4 mol O₃;
 2 mol H₂O:3 mol CO₂; or their reciprocals
 - b. C_3H_4 is 0.5X; O_2 is 2X; and CO_3 is 1.5X

SECTION 2

PROBLEMS

- 1. 4.5 mol
- 2. 200 gt
- 3. 0.53 g
- 4. 34.8 g
- 5. a. 60.2 g
 - b. 42.1 L
- 6. a. 81 g.
 - b. 2.9 mol
 - c. $1.3 \times 10^{2} \, \text{g}$

SECTION 3

PROBLEMS

- 1. 88%
- 2. a. N2: 2.0 mol.
 - b. 8.0 mol.
 - c. 6.4 mod
- 3. a. 0.10 mol
 - b. HCL
 - e. 1.4 m.
- 4. m. 1.26 × 10³ gt
 - b. 960. g
 - $c. 6.9 \times 10^{3} L$

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Tobias Bleicker

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