

**Atomic Structure & Theory**

1. Can the number of protons in an element ever change?

- ☐ A. No, protons and neutrons are bonded together in atomic nuclei, and neither can change.
  - ☐ B. Yes, isotopes of the same element have different numbers of protons.
  - ☐ C. No, if the number of protons changes, the element changes.
  - ☐ D. Yes, when different elements bond, they can share or transfer protons.
- 

**Atomic Structure & Theory**

2. All matter, both living and nonliving, is made up of tiny particles called \_\_\_\_\_.

- ☐ A. cells
  - ☐ B. atoms
  - ☐ C. dust
  - ☐ D. bacteria
- 

**The Periodic Table**

4. Which of the following is true of all the elements in group 2 on the periodic table?

- I. They are all metals.
  - II. They all have two electrons in their outermost shell.
  - III. They all tend to lose electrons and become positive ions.
  - IV. They all have similar chemical properties.
- ☐ A. II, III, and IV only
  - ☐ B. I, III, and IV only
  - ☐ C. I, II, III, and IV
  - ☐ D. III and IV only
-

# Chemistry Semester 1 Exam Review Study Island

**LP Steffe**



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