

Chemical vs. Physical Change Experiment



Chemical vs. Physical Change Experiment

Name: _____

Objective

To observe and compare the reactions of vinegar with baking soda and water with vinegar to determine if a chemical or physical change occurs.

Materials

- 2 small cups
- Vinegar
- Baking soda
- Water
- Spoon
- Observing station
- Worksheet for observations

Hypothesis

1. What do you think will happen when vinegar is mixed with baking soda?

2. What do you think will happen when vinegar is mixed with water?

Procedure

1. Pour two tablespoons of vinegar into each cup.
2. In the first cup, add one teaspoon of baking soda and observe.
3. In the second cup, add one teaspoon of water and observe.
4. Record your observations below.

Answer Key: Chemical vs. Physical Change Experiment

Hypothesis (answers will vary, but example responses)

1. When vinegar is mixed with baking soda, I think it will bubble or fizz.

Answer: When vinegar is mixed with water, I think it will just mix together without any fizz.

Yes

yes = Baking Soda: There were bubbles forming and gas formed. It looked like a volcano or volcano.

yes = Water: The liquids just mixed. Nothing really happened.

Are bubbles/bubbles or gas formed?

- In the reaction with baking soda, it bubbled and fizzed.

Other reactions create a new substance? How do you know?

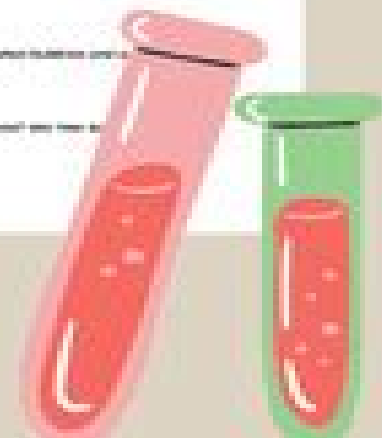
- The reaction with baking soda made a gas (carbon dioxide). That means a new substance was created.

Are bubbles new or chemical change? Why?

- reaction between vinegar and baking soda. It created bubbles and it is chemical change happened.

Are bubbles new or physical change? Why?

- vinegar and water. They just mixed together without any new substance.



Chemistry 130 Experiment 3 Physical And Chemical Change

Xiaolong Qi



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