

Question 2

- 2(b) The zinc decreases in size, the blue colour of the solution fades and a red-brown solid is deposited in the test tube and/or on the solid zinc.
- 2(c) Add the suspected reducing agent to a few cm³ of an oxidizing agent such as H⁺_(aq)/KMnO_{4(aq)} in a test tube. If the KMnO₄ changes from purple to colourless, the presence of a reducing agent is confirmed.

Question 3

- 3(a) Members of a homologous series:
- Have the same general formula.
 - Show similar chemical reactions though varying in vigour.
 - Differ in molecular formula from the next by CH₂.

3(b)(i)



Compound A

3(b)(ii) Compound A belongs to the alkanes.

3(b)(iii) C₃H₁₂.

3(c)(i)



Compound B

3(c)(ii) Compound B belongs to the alkenes.

3(c)(iii) C₃H₁₀.

3(d)(i) Carboxylic acid: C_nH_{2n+1}COOH where n ≥ 0.

Alcohol: C_nH_{2n+1}OH where n ≥ 1.

3(d)(ii) Carboxylic acid: methanoic acid.

Alcohol: methanol.

Chemistry Cxc Past Papers And Answers

Arthur James Wells



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