#### Question 2

- 2(b) The zinc decreases in size, the blue colour of the solution fades and a red-brown solid is deposited in the test tube and/or on the solid zinc.
- 2(c) Add the suspected reducing agent to a few cm<sup>3</sup> of an oxidizing agent such as H\*<sub>(ac)</sub>/KMnO<sub>4(ac)</sub> in a test tube. If the KMnO<sup>4</sup> changes from purple to colourless, the presence of a reducing agent is confirmed.

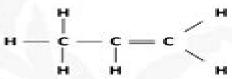
#### Question 3

- 3(a) Members of a homologous series:
  - Have the same general formula.
  - Show similar chemical reactions though varying in vigour.
  - Differ in molecular formula from the next by CH<sub>2</sub>.
- 3(b)(ii)



#### Compound A

- 3(b)(ii) Compound A belongs to the alkanes.
- 3(b)(iii) C<sub>5</sub>H<sub>12</sub>.
- 3(c)(i)



Compound B

- 3(c)(ii) Compound B belongs to the alkenes.
- 3(c)(iii) C<sub>5</sub>H<sub>10</sub>-
- 3(d)(i) Carboxylic acid:  $C_nH_{2n-1}COOH$  where  $n \ge 0$ .
  - Alcohol: C,H2m-1OH where n ≥ 1.
- 3(d)(iii) Carboxylic acid: methanoic acid.

Alcohol: methanol.

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**Arthur James Wells** 

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