# Chemistry of Life notes

Organic Molecules: molecules that contain both Carbon and Hydrogen (C and H); can contain
other elements too; - IF THEY CONTAIN ONLY C AND H then they can ALSO be called
hydrocarbons.

What is so cool about carbon?

Carbon has \_four valence electrons which means it can form 4 bonds AND can bond with itself.

- This means it can form long chains, rings, branched chains and many other molecules

Inorganic molecules: molecules that do NOT contain both C and H







WATER: most essential inorganic molecule

- ♦ Body is 65- 75% water on average
- More water in fat or muscle?? MUSCLE Why? Fat is hydrophobic
- More water in Males or females?? Males Why? Males have less fat than females

#### Functions of water:

- Solvent: dissolves lots of stuff (salts especially)
- Medium: where the chemical reactions happen
- Moistens Surfaces: lungs, mouth, nasal passages
- Temperature Regulation: an example is sweating (keeps internal temperature from getting too high)
- Cushion Brain in skull;
- Transportation: moves nutrients and other materials through the body (blood is mostly water!)
- · Lubrication: in your Joints; prevent bones from scraping
- Hydrolysis: type of chemical reaction that uses water to break apart large molecules

# **Chemistry Of Life Lecture Notes Jdenuno**

**Meeting on Chemistry and Life** 

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The Chemistry of Life's Origins | Mayo Greenberg, CX Mendoza-Gomez, Valerio Pirronello, 1993-10-31 Lectures on the Chemistry of Life Thomas Turner, Frederick Crace Calvert, 1847 What is Life? Addy Pross, 2012-09-27 Pross examines these issues from a chemical perspective providing a new understanding of how the sciences of chemistry and The Chemistry of Life Martin Olomucki, 1993 The Tao of Chemistry and Life Eugene H. biology relate to one another Cordes, 2009-06-04 Written with the non scientist in mind this book employs the molecule and its interactions to explain the characteristics of living organisms in terms of the underlying chemistry of life Following introductory chapters on the fundamentals of life attention then turns to small molecules such as hormones and neurotransmitters and subsequently to macromolecules including proteins and nucleic acids The interactions between small and macromolecules remains a central point throughout the book These include enzymatic catalysis hormone action neurotransmission regulation of metabolism biosynthesis of macromolecules the mechanism of action of drugs taste olfaction learning and memory and chemical communication A second central point of emphasis is the sensitive relationship between chemical structure and biological activity Examples abound and include why subtle changes in fatty acid architecture have positive or negative outcomes for human health in omega three fatty acids and trans fats and how modest changes in the chemical decoration of the steroid skeleton provide the difference between male and female sex hormones Beyond these examples taken from the chemistry of small molecules the book includes a thoughtful consideration of genomics including the relationship between genome structure and species The theme of human health appears throughout the book Cardiovascular medicine cancer metabolic diseases and diseases of the nervous system receive significant attention including consideration of how a variety of drugs work in combating these issues In sum the goal of this book is to inform the non scientist community in a way that will lead to increased understanding of the relationship between chemistry and life The Chemistry of Life Robert Thornton, 2001

**Chemistry and Life** Meeting on Chemistry and Life,2008 **Chemistry of Life** Peter P. Berlow, 1982 Chemistry of Life Katherine Blood Hoffman, 1964 The Chemistry of Life John Carlton Williams, 1965 The Chemistry of Life Joseph Needham, 1970 The Physics and Chemistry of Life Scientific American, inc,1957 The Chemistry of Human Life George W. Carey, 1996-09 Chemistry of Life and Cells II Laying the Foundation, 2009-07 Chemistry of Life Standalone Benjamin Cummings Staff, 1998-01-01 Lectures Notes for a Course in Biological Chemistry Lawrence Joseph Henderson, Chemistry and Life Oldrich Zmeskal, 2019-05-27 7th Meeting on Chemistry and Life 2018 Selected peer reviewed papers from the 7th Meeting on Chemistry and Life 2018 Brno Czech Republic Chemistry of Life ,2025-10-14 **Chemistry of Life** Saunochem, 1995-01-01 Physics and Chemistry of Life Simon, 1955

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