

Inorganic Chemistry

Transition Metals

General Properties

Transition metals are elements found in the d-block of the periodic table, characterized by their ability to form colored compounds and complex ions. They exhibit a range of oxidation states and are known for their catalytic activity.

The first row of transition metals includes Scandium (Sc) through Nickel (Ni). These elements are typically found in the Earth's crust and are used in various industrial applications, such as in the production of alloys and catalysts.

The second row of transition metals includes Titanium (Ti) through Manganese (Mn). These elements are also found in the Earth's crust and are used in various industrial applications, such as in the production of alloys and catalysts.

The third row of transition metals includes Zirconium (Zr) through Cadmium (Cd). These elements are also found in the Earth's crust and are used in various industrial applications, such as in the production of alloys and catalysts.

The fourth row of transition metals includes Hafnium (Hf) through Mercury (Hg). These elements are also found in the Earth's crust and are used in various industrial applications, such as in the production of alloys and catalysts.

Coordination Chemistry

Coordination Compounds

Coordination compounds are formed when a central metal ion is surrounded by ligands. The coordination number is the number of ligands bonded to the central metal ion.

Organometallic Chemistry

Organometallic Compounds

Organometallic compounds are formed when a metal is bonded to an organic group. They are used in various industrial applications, such as in the production of polymers and catalysts.

Supramolecular Chemistry

Supramolecular Compounds

Supramolecular compounds are formed by the assembly of molecules through non-covalent interactions. They are used in various industrial applications, such as in the production of materials and catalysts.

Environmental Chemistry

Environmental Compounds

Environmental compounds are formed by the interaction of inorganic and organic substances in the environment. They are used in various industrial applications, such as in the production of materials and catalysts.

Physical Chemistry

Physical Compounds

Physical compounds are formed by the interaction of inorganic and organic substances in the environment. They are used in various industrial applications, such as in the production of materials and catalysts.

Biological Chemistry

Biological Compounds

Biological compounds are formed by the interaction of inorganic and organic substances in the environment. They are used in various industrial applications, such as in the production of materials and catalysts.

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B Lingard



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Inorganic Chemistry Mark Jackson, 2010-05-31 Our 3 panel 6 page guide is jam packed with chemistry information not covered in organic chemistry and biochemistry Perfect for science students who want to focus on the aspects of chemistry that go beyond what is covered in material that deals with organic topics this guide covers the key concepts principles figures and formulas that inorganic chemistry students will need to know in order to succeed Tables images and graphic elements further enhance the text

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