



## illuminati - 2019

## Advanced Chemistry Assignment-3B | Inorganic Chemistry

## SINGLE CHOICE QUESTIONS

- During refining of pig iron by bessemerization process, which of the following impurity is oxidized last.  
(A) Si (B) Mn (C) S (D) C
- $\text{PCl}_5 + \text{H}_2(\text{g}) \xrightarrow{\text{warm}} \text{X} + \text{Y}$   
If molecular weight of 'X' is greater than that of 'Y' then compound 'X' reacts with :  
(A)  $\text{H}_2\text{O}(\text{g})$  (B)  $\text{Cl}_2$  gas (C)  $\text{O}_2$  gas (D) All
- Incorrect statement for borax is :  
(A) Its aqueous solution acts as buffer  
(B) Its aqueous solution is used to determine strength of dil. HCl using phenolphthalein.  
(C) It is used as acidic flux in brazing  
(D) H-bonding is present in borax
- $\text{H}_2\text{O}_2$  acts as oxidizing agent with which of the following reagent :  
(A)  $\text{KMnO}_4$  in basic medium (B)  $\text{Fe}_2(\text{SO}_4)_3$   
(C)  $\text{I}_2$  in alkaline medium (D)  $\text{HClO}_4$
- Milk of lime +  $\text{SOCl}_2 \longrightarrow \text{P} \downarrow + \text{Q}$   
Incorrect statement for compound 'P' and 'Q' is :  
(A) 'Q' produces observable change with hypo  
(B) 'P' dissolves in excess of  $\text{H}_2\text{SO}_4$   
(C) 'Q' produces precipitate with aqueous solution of KPF  
(D) 'P' reacts with  $\text{H}_2\text{S}$  gas in the presence of dilHCl
- $\text{K}_4[\text{Fe}(\text{CN})_6] + \text{conc. H}_2\text{SO}_4 \xrightarrow{\text{warm}} \text{products}$   
Incorrect statement for products is :  
(A) Gaseous product is used in Mond's process  
(B) Oxidation state of iron remains same  
(C) One of the cations is derived from nonmetals  
(D) Gaseous product obtained is acidic
- $3\text{HgO} \cdot \text{HgCO}_3(\text{s}) + 2\text{PbS}(\text{s}) \xrightarrow{\Delta} \text{Products}$   
Incorrect statements about products is :  
(A) One of the liberated gas can decolourize aqueous  $\text{KMnO}_4$  solution.  
(B) Mercury is obtained in elemental state  
(C) Lead is obtained in the form of  $\text{PbO}_2$   
(D)  $\text{PbS}$  is roasted into  $\text{PbO}$  and  $\text{SO}_2$
- Incorrect statement for complex ions  $[\text{Mn}(\text{H}_2\text{O})_6]^{2+}$  and  $[\text{MnBr}_4]^{2-}$  is :  
(A) Intensity of colour in  $[\text{MnBr}_4]^{2-}$  is less than  $[\text{Mn}(\text{H}_2\text{O})_6]^{2+}$   
(B) Both have same magnetic moment  
(C) Stability :  $[\text{Mn}(\text{H}_2\text{O})_6]^{2+} > [\text{MnBr}_4]^{2-}$   
(D) Valence shell d- orbitals of  $\text{Mn}^{2+}$  are used in its hybridization in  $[\text{Mn}(\text{H}_2\text{O})_6]^{2+}$

# Inorganic Chemistry Questions And Answers Pdf

**S. D. Titmas**



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