

Chapter 2 Review Questions

1. Elements are pure substances that cannot be broken down into simpler substances.
2. Three things are pencils, coins, and a portable music player.
3. a) P=Phosphorus b) Be=Beryllium c) K=Potassium d) Co=Cobalt
4. a) Helium=He b) Lithium=Li c) Boron=B d) Magnesium=Mg e) Calcium=Ca
5. Two elements are: mercury, and bromine.
6. Metals are ductile, malleable, a solid, shiny, good conductor of heat, and electricity, and they lose electrons, and become positive during reactions.
7. a) An element is called carbon which is stronger than steel.
b) It is called steel.
8. It has a heavier atomic mass, and mercury turns into liquid at room temperature.
9. The chart is called a periodic table.
10. It measures the average atom of an element.
11. The atomic number is the same number as the number protons.
12. Is elements with similar physical and chemical properties of other elements.
13. One and Two are both gases, and they are colourless. Seventeen and eighteen are both gases at room temperature only fluorine, and chlorine, and they are non-metals.
14. They react with water, but not as vigorously as normal alkali metals do.
15. The most important property is the most stable, and unreactive elements.
16. It represents the number of electrons there should be around the atom which makes it either a atom or a ion.
17. a) Is how many shells of electrons are around a atom or a ion they go in different pairs or sets.
b) Valence electrons are how many electrons are around the outermost layer of the atom.
18. a) Potassium=1 b) Aluminum=3 c) Oxygen=6 d) Argon=8
19. a) Noble gases is the chemical family with filled valence shells.
b) Having filled shells makes it stable so they will be less reactive.
20. A chemical property is a chemical or substance that undergoes a chemical change or reaction, and also they are properties that do not change the chemical nature of matter. A physical change is properties do not change the chemical nature of matter.
21. The atomic number goes up by one in a pattern from left to right and to left again, and there are mixed all in different groups.
22. a) Antimony=51 b) Arsenic=33 c) Manganese=25 d) Selenium=34
23. Hydrogen=1.0 Oxygen=16.0 Nitrogen=14.0 Rhenium=186.2
a) Rhenium is the heaviest of 186.2 amu
b) Hydrogen is the lightest of 1.0 amu
24. a) Germanium, and Neptunium does of +4.
b) Neptunium does of +3,+4,+5,+6.
c) Nickel, and Neodymium are metalloids.

Chapter 2 Review Questions And Answers

Jeff Doyle, Jennifer DeHaven Carroll



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Routing TCP/IP Jeff Doyle, Jennifer DeHaven Carroll, 2001 Intended for courses in TCP IP routing protocols and advanced networking This volume presents an examination of exterior routing protocols EGP and BGP and advanced IP routing issues such as multicast routing quality of service routing Ipv6 and router management It enables students learn IP design and management techniques

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Customizing AutoCAD 2004 Sham Tickoo,CADCIM Technologies,2004 Written in accordance with the design capabilities of AutoCAD 2004 this updated edition offers detailed explanations of customizing techniques for advanced users of AutoCAD All the various levels of customization in AutoCAD are examined in one comprehensive volume from the basic topics of creating template drawings and customizing menus to the more advanced features such as modifying the AutoCAD environment in ways that help industry professionals meet the needs of their organization Thorough explanations are enhanced by live projects and examples that make it easy to comprehend and master the customizing concepts of AutoCAD 2004 *Pretreatment Facility Inspection* ,1988

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