

Chapter 3 – Stoichiometry

Exercise 3.1: Determine the molecular/molar masses of the following molecules:

- | | |
|--|----------------------------------|
| a. H_2O 18.015 | d. NaOH 39.997 |
| b. $\text{Sulfuric acid (H}_2\text{SO}_4\text{)}$ 98.079 | e. H_2O_2 34.015 |
| c. C_2H_6 30.070 | f. CO_2 44.009 |

Exercise 3.2: Determine the molecular/molar masses of the following molecules:

- | | |
|-----------------------------------|----------------------------------|
| a. H_2SO_4 98.079 | d. H_2O 18.015 |
| b. NaOH 39.997 | e. NaOH 39.997 |
| c. NaOH 39.997 | f. Sulfuric acid 98.079 |

Exercise 3.3: Express the following grams of each component in moles of each component:

- | | |
|---|---|
| a. 27.0 kg C_2H_6 1.000 mol | d. 1.0 kg Sulfuric acid 0.0102 mol |
| b. 1.0 kg Sulfuric acid 0.0102 mol | e. 0.01 kg H_2O 0.555 mol |
| c. 0.01 kg C_2H_6 0.333 mol | f. 0.01 kg CO_2 0.227 mol |

Exercise 3.4: Express the following grams of each component in moles of each component:

- | | |
|---|--|
| a. 0.01 kg H_2O_2 0.333 mol | d. 0.01 kg H_2O 0.555 mol |
| b. 0.01 kg H_2O_2 0.333 mol | e. 0.01 kg Sulfuric acid 0.0102 mol |
| c. 0.01 kg H_2O_2 0.333 mol | f. 0.01 kg Sulfuric acid 0.0102 mol |

Exercise 3.5: Express the following grams of each component in moles of each component:

- | | |
|---|--|
| a. 0.01 kg H_2O_2 0.333 mol | d. 0.01 kg Sulfuric acid 0.0102 mol |
| b. 0.01 kg C_2H_6 0.333 mol | e. 0.01 kg H_2O 0.555 mol |
| c. 0.01 kg C_2H_6 0.333 mol | f. 0.01 kg CO_2 0.227 mol |

Exercise 3.6: Express the following grams of each component in moles of each component:

- | | |
|---|--|
| a. 0.01 kg H_2O_2 0.333 mol | d. 0.01 kg H_2O 0.555 mol |
| b. 0.01 kg H_2O_2 0.333 mol | e. 0.01 kg Sulfuric acid 0.0102 mol |
| c. 0.01 kg H_2O_2 0.333 mol | f. 0.01 kg Sulfuric acid 0.0102 mol |

Exercise 3.7: Determine the following chemical conditions:

- | | |
|-------------------------------------|-------------------------------------|
| a. H_2O_2 0.333 mol | d. H_2O_2 0.333 mol |
| b. C_2H_6 0.333 mol | e. H_2O_2 0.333 mol |
| c. H_2O_2 0.333 mol | f. H_2O_2 0.333 mol |

Chapter 3 Stoichiometry Exercises Answers

Steven S. Zumdahl, Kenneth C. Brooks



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