

Chemistry 120 – Chapter 3 – Matter

- Matter
 - o Has mass, takes up space
 - o Has four states: Gas, Liquid, Solid, [Plasma, not gone over in intro course]
- Atoms and Molecules
 - o Fundamental unit of matter is the atom. Atoms are particles that are incredibly small and comes in different varieties
 - o Molecules are groups of two or more atoms held together by covalent bonds
 - contain a well-defined number of atoms that is expressed in a whole number ratio
- States of Matter
 - o Which state a substance is depends on two factors: Temperature and Pressure
 - o Gases: takes the shape of their container and completely fill their container
 - Gases are composed of molecules that are far apart from one another and travel in random paths
 - o Liquids: takes the shape of their container and do not fill their container completely
 - Liquid particles freely move across one another
 - o Solids: Do not take the shape of their container and do not fill their container completely
 - At the submicroscopic level, we see two types of solids
 - Crystalline: composed of atoms in highly-ordered arrangements like table salt and diamonds
 - Amorphous: much less organization among atoms and appear to be random in some cases like Styrofoam, rubber, glass, and plastics
- Classification of Matter
 - o Classified as either pure substance or mixture
 - Pure substances include elements and compounds
 - Pure substances have a consistent, uniform composition
 - Elements are made up solely of the same type of atom
 - Compounds are made up of atoms in a fixed whole number ratio
 - Mixtures include homogeneous or heterogeneous
 - Composition is not fixed
 - Homogeneous mixtures have all parts in the same state and all parts must be mixed together
 - Heterogeneous mixtures are simply not homogeneous
- Separations
 - o Mixtures can be separated by physical methods
 - o Compounds can only be separated into individual elements by chemical reactions
- Properties of Matter
 - o Matter can be described in two ways: Chemical Properties and Physical Properties
 - Chemical properties describe how a type of matter interacts or reacts with another type of matter
 - Physical properties which include all non-chemical properties
- Changes of Matter
 - o Can be described the same way as properties with chemical and physical changes

Chapter 6 Study Guide Chemistry

Ying-Ying Zheng



Chapter 6 Study Guide Chemistry:

This book delves into Chapter 6 Study Guide Chemistry. Chapter 6 Study Guide Chemistry is a crucial topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Chapter 6 Study Guide Chemistry, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Chapter 6 Study Guide Chemistry
 - Chapter 2: Essential Elements of Chapter 6 Study Guide Chemistry
 - Chapter 3: Chapter 6 Study Guide Chemistry in Everyday Life
 - Chapter 4: Chapter 6 Study Guide Chemistry in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, this book will provide an overview of Chapter 6 Study Guide Chemistry. The first chapter will explore what Chapter 6 Study Guide Chemistry is, why Chapter 6 Study Guide Chemistry is vital, and how to effectively learn about Chapter 6 Study Guide Chemistry.
 3. In chapter 2, the author will delve into the foundational concepts of Chapter 6 Study Guide Chemistry. The second chapter will elucidate the essential principles that need to be understood to grasp Chapter 6 Study Guide Chemistry in its entirety.
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 5. In chapter 4, this book will scrutinize the relevance of Chapter 6 Study Guide Chemistry in specific contexts. This chapter will explore how Chapter 6 Study Guide Chemistry is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, this book will draw a conclusion about Chapter 6 Study Guide Chemistry. The final chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Chapter 6 Study Guide Chemistry.

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