

CHAPTER 7 REVIEW*Chemical Formulas and Chemical Compounds***SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

1. C In a Stock name such as iron(III) sulfate, the roman numeral tells us _____
(a) how many atoms of Fe are in one formula unit
(b) how many sulfate ions can be attached to the iron atom
(c) the charge on each Fe ion
(d) the total positive charge of the formula unit
2. C The result of changing a subscript in a correctly written chemical formula is to _____
(a) change the number of moles represented by the formula
(b) change the charges on the other ions in the compound
(c) change the formula so that it no longer represents the compound it previously represented
(d) have no effect on the formula
3. The explosive TNT has the molecular formula $C_7H_5(NO_2)_3$.
4 elements a. How many elements make up this compound?
6 oxygen atoms b. How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$?
21 atoms c. How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$?
 4.2×10^{24} atoms d. How many atoms are present in a sample of 2×10^{23} molecules of $C_7H_5(NO_2)_3$?
4. How many atoms are present in each of these formula units?
11 atoms a. $Ca(HCO_3)_2$
45 atoms b. $C_{12}H_{22}O_{11}$
10 atoms c. $Fe(ClO_2)_3$
9 atoms d. $Fe(ClO_3)_2$
5. N_2O_5 a. What is the formula for the compound dinitrogen pentoxide?
carbon(IV) sulfide b. What is the Stock name for the covalent compound CS_2 ?
 H_2SO_3 c. What is the formula for sulfurous acid?
phosphoric acid d. What is the name for the acid H_3PO_4 ?

Chapter 7 Section 3 Modern Chemistry Review Answers

Patrick Vollmar



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Table of Contents Chapter 7 Section 3 Modern Chemistry Review Answers

1. Understanding the eBook Chapter 7 Section 3 Modern Chemistry Review Answers
 - The Rise of Digital Reading Chapter 7 Section 3 Modern Chemistry Review Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 7 Section 3 Modern Chemistry Review Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 7 Section 3 Modern Chemistry Review Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 7 Section 3 Modern Chemistry Review Answers
 - Personalized Recommendations
 - Chapter 7 Section 3 Modern Chemistry Review Answers User Reviews and Ratings

- Chapter 7 Section 3 Modern Chemistry Review Answers and Bestseller Lists
- 5. Accessing Chapter 7 Section 3 Modern Chemistry Review Answers Free and Paid eBooks
 - Chapter 7 Section 3 Modern Chemistry Review Answers Public Domain eBooks
 - Chapter 7 Section 3 Modern Chemistry Review Answers eBook Subscription Services
 - Chapter 7 Section 3 Modern Chemistry Review Answers Budget-Friendly Options
- 6. Navigating Chapter 7 Section 3 Modern Chemistry Review Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 7 Section 3 Modern Chemistry Review Answers Compatibility with Devices
 - Chapter 7 Section 3 Modern Chemistry Review Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 7 Section 3 Modern Chemistry Review Answers
 - Highlighting and Note-Taking Chapter 7 Section 3 Modern Chemistry Review Answers
 - Interactive Elements Chapter 7 Section 3 Modern Chemistry Review Answers
- 8. Staying Engaged with Chapter 7 Section 3 Modern Chemistry Review Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 7 Section 3 Modern Chemistry Review Answers
- 9. Balancing eBooks and Physical Books Chapter 7 Section 3 Modern Chemistry Review Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 7 Section 3 Modern Chemistry Review Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 7 Section 3 Modern Chemistry Review Answers
 - Setting Reading Goals Chapter 7 Section 3 Modern Chemistry Review Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 7 Section 3 Modern Chemistry Review Answers
 - Fact-Checking eBook Content of Chapter 7 Section 3 Modern Chemistry Review Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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