

Section 6.2

Covalent Bonding

◉ Main Ideas:

- Dots placed around an element's symbol can represent valence electrons
- **Electron-dot notations** can represent compounds.
- Some atoms can share multiple pairs of electrons.

Chapter 6 Chemical Bonds Section 6 2 Covalent Bonding

Roman Wölfel



Chapter 6 Chemical Bonds Section 6 2 Covalent Bonding:

Chapter 6 Chemical Bonds Section 6 2 Covalent Bonding Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the energy of words has been evident than ever. They have the capacity to inspire, provoke, and ignite change. Such is the essence of the book **Chapter 6 Chemical Bonds Section 6 2 Covalent Bonding**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book's key themes, examine its writing style, and analyze its overall impact on readers.

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