

CHAPTER 7 REVIEW*Chemical Formulas and Chemical Compounds***SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

- 1.** C In a Stock name such as iron(III) sulfate, the roman numeral tells us _____.
 - (a) how many atoms of Fe are in one formula unit
 - (b) how many sulfate ions can be attached to the iron atom
 - (c) the charge on each Fe ion
 - (d) the total positive charge of the formula unit
- 2.** C The result of changing a subscript in a correctly written chemical formula is to _____.
 - (a) change the number of moles represented by the formula
 - (b) change the charges on the other ions in the compound
 - (c) change the formula so that it no longer represents the compound it previously represented
 - (d) have no effect on the formula
- 3.** The explosive TNT has the molecular formula $C_7H_5(NO_2)_3$.

<u>4 elements</u>	a. How many elements make up this compound?
<u>6 oxygen atoms</u>	b. How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$?
<u>21 atoms</u>	c. How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$?
<u>4.2×10^{24} atoms</u>	d. How many atoms are present in a sample of 2×10^{23} molecules of $C_7H_5(NO_2)_3$?
- 4.** How many atoms are present in each of these formula units?

<u>11 atoms</u>	a. $Ca(HCO_3)_2$
<u>45 atoms</u>	b. $C_{12}H_{22}O_{11}$
<u>10 atoms</u>	c. $Fe(ClO_2)_3$
<u>9 atoms</u>	d. $Fe(ClO_3)_3$
- 5.**

<u>N_2O_5</u>	a. What is the formula for the compound dinitrogen pentoxide?
<u>carbon(IV) sulfide</u>	b. What is the Stock name for the covalent compound CS_2 ?
<u>H_2SO_3</u>	c. What is the formula for sulfurous acid?
<u>phosphoric acid</u>	d. What is the name for the acid H_3PO_4 ?

Chapter 7 Review Chemical Formulas And Chemical Compounds

DJ Losen



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Table of Contents Chapter 7 Review Chemical Formulas And Chemical Compounds

1. Understanding the eBook Chapter 7 Review Chemical Formulas And Chemical Compounds
 - The Rise of Digital Reading Chapter 7 Review Chemical Formulas And Chemical Compounds
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 7 Review Chemical Formulas And Chemical Compounds
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 7 Review Chemical Formulas And Chemical Compounds
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 7 Review Chemical Formulas And Chemical Compounds
 - Personalized Recommendations
 - Chapter 7 Review Chemical Formulas And Chemical Compounds User Reviews and Ratings
 - Chapter 7 Review Chemical Formulas And Chemical Compounds and Bestseller Lists

5. Accessing Chapter 7 Review Chemical Formulas And Chemical Compounds Free and Paid eBooks
 - Chapter 7 Review Chemical Formulas And Chemical Compounds Public Domain eBooks
 - Chapter 7 Review Chemical Formulas And Chemical Compounds eBook Subscription Services
 - Chapter 7 Review Chemical Formulas And Chemical Compounds Budget-Friendly Options
6. Navigating Chapter 7 Review Chemical Formulas And Chemical Compounds eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 7 Review Chemical Formulas And Chemical Compounds Compatibility with Devices
 - Chapter 7 Review Chemical Formulas And Chemical Compounds Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 7 Review Chemical Formulas And Chemical Compounds
 - Highlighting and Note-Taking Chapter 7 Review Chemical Formulas And Chemical Compounds
 - Interactive Elements Chapter 7 Review Chemical Formulas And Chemical Compounds
8. Staying Engaged with Chapter 7 Review Chemical Formulas And Chemical Compounds
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 7 Review Chemical Formulas And Chemical Compounds
9. Balancing eBooks and Physical Books Chapter 7 Review Chemical Formulas And Chemical Compounds
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 7 Review Chemical Formulas And Chemical Compounds
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chapter 7 Review Chemical Formulas And Chemical Compounds
 - Setting Reading Goals Chapter 7 Review Chemical Formulas And Chemical Compounds
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter 7 Review Chemical Formulas And Chemical Compounds
 - Fact-Checking eBook Content of Chapter 7 Review Chemical Formulas And Chemical Compounds
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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