

Chapter Test B

Chapter: Chemical Equations and Reactions

PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. The production of a slightly soluble solid compound in a double-displacement reaction results in the formation of a
a. gas.
b. precipitate.
c. combustion reaction.
d. halogen.
- _____ 2. To balance a chemical equation, it is permissible to adjust the
a. coefficients.
b. subscripts.
c. formulas of the products.
d. number of products.
- _____ 3. In a chemical equation, the symbol (aq) indicates that the substance is
a. water.
b. dissolved in water.
c. an acid.
d. insoluble.
- _____ 4. The tendency for a replacement reaction to occur increases as the
a. interval between any two elements in the activity series decreases.
b. temperature decreases.
c. valence electrons are used up.
d. interval between any two elements in the activity series increases.
- _____ 5. The coefficients in a chemical equation
a. indicate the number of moles of each substance that react.
b. show the number of grams of each substance that react.
c. are the molar masses of the substances.
d. show the valence electrons for each atom.
- _____ 6. If metal X is lower than metal Y in the activity series, then
a. X will replace ions of Y in solution.
b. Y will replace ions of X in solution.
c. Y will form oxides only indirectly.
d. X will react with cold water.

Chapter 8 Test B

Jay K. Hackett



Chapter 8 Test B:

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