

9

CHEMICAL NAMES AND FORMULAS

Chapter Test A

A. Matching

Match each description in Column B to the correct term in Column A. Write the letter of the correct description in the blank.

Column A

- _____ 1. cation
- _____ 2. anion
- _____ 3. law of definite proportions
- _____ 4. acid
- _____ 5. base
- _____ 6. law of multiple proportions
- _____ 7. polyatomic ion
- _____ 8. monatomic ion
- _____ 9. binary compound

Column B

- a. any atom or group of atoms with a negative charge
- b. When two elements combine to form more than one compound, the different masses of one element that combine with the same mass of the other element are in the ratio of small whole numbers.
- c. tightly bound group of atoms that behaves as a unit and carries a charge
- d. a compound that produces hydroxide ions when dissolved in water
- e. a compound composed of two different elements
- f. any atom or group of atoms with a positive charge
- g. In all samples of the same chemical compound, the elements are always combined in the same proportion by mass.
- h. a compound that produces hydrogen ions when dissolved in water
- i. consists of a single atom with a positive or negative charge

B. Multiple Choice

Choose the best answer and write its letter on the line.

- _____ 10. The correct name for the N^{3-} ion is the:
 - a. nitrate ion.
 - b. nitric ion.
 - c. nitride ion.
 - d. nitrite ion.

Chapter 9 Chemical Names And Formulas Test B Answers

M. Scovell



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