

14

THE BEHAVIOR OF GASES

Chapter Test II

A. Matching

Match a word or phrase in Column I with the correct description in Column II. Write the letter of the correct description next to the item.

Column I

1. The constant volume with temperature, the heat capacity of a substance per unit temperature rise of the substance, or the amount of heat necessary to raise the temperature of 1 mole of a substance 1°C
2. The volume of a fixed amount of a gas at constant pressure and temperature
3. The volume of a gas at constant pressure and temperature, proportional to the number of moles of gas
4. The volume of a gas at constant pressure and temperature, proportional to the absolute temperature
5. A constant of a gas that is the product of the gas constant and the number of moles of gas
6. The pressure of a gas at constant volume, proportional to the absolute temperature
7. The volume of a gas at constant pressure and temperature, proportional to the absolute temperature
8. $\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$
9. The volume of a gas at constant pressure and temperature, proportional to the absolute temperature
10. A constant of a gas that is the product of the gas constant and the number of moles of gas

Column II

- a. molar heat capacity
- b. ideal gas constant R
- c. volume
- d. compressibility
- e. Boyle's law
- f. Boyle's constant
- g. Charles' law of expansion
- h. volume
- i. compressibility
- j. Charles' law of expansion

B. Multiple Choice

Choose the correct answer and write its letter on the line.

1. Boyle's law states that the volume of a fixed amount of gas is inversely proportional to the pressure, if the temperature is constant. This is true for all gases.
- a. true for all gases
- b. true for all gases
- c. true for all gases
- d. true for all gases

Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers

Ian Pickup



Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers:

Chemistry and Our Universe , In the first of two lectures on the properties of gases review the basic equations that describe their behavior Learn the history of Boyle s law Gay Lussac s law Charles s law and Avogadro s law Then use these four expressions to derive the celebrated ideal gas law

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