

Chemistry Worksheet Class 6 on Chapter 6 Changes Around Us - Set 1

Q1. Which of the following is a reversible change?

- a.) Formation of curd.
- b.) Ripening of fruits.
- c.) Evaporation.
- d.) None of the above.

Q2. Irreversible changes are-

- a.) Permanent
- b.) Temporary
- c.) Physical
- d.) Chemical

Q3. Water may change to-

- a.) Ice
- b.) Water vapour
- c.) Both (a.) and (b.)
- d.) None of the above

Q4. Fill in the blanks.

- a.) The changes that can be reversed are called ____.
- b.) On heating, the metals ____.
- c.) Water changes into ____ due to heating.

Q5. State True or False.

- a.) Ripening of fruits is a reversible change.
- b.) The burning of candles is an irreversible change.
- c.) Dissolving salt in water is a reversible change.

Q6. Define Evaporation.

Q7. Define fast and slow changes.

Q8. Explain why most physical changes are reversible.

Chemistry Matter And Change Chapter 6 Answers

McGraw-Hill Staff



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Chemistry: Molecules, Matter, and Change Media Activities Book Loretta Jones, Carl Hoeger, Peter William Atkins, Regina Schoenfeld-Tacher, 2000-01-15 The Media Activity Book MAB for Jones Atkins Chemistry molecules matter and change contains chapters with lists and descriptions of some of the media available as you study the chapter Each activity begin with a specific textbook reference Then you are given a time estimate of how long it will take to use the media An M media icon in the margin of the textbook means that media exists to support that area of text The media is found in three different places on the website and on two CDs **Chemistry: Matter & Change, Solving Problems - A Chemistry Handbook** McGraw Hill, 2001-08 Glencoe Chemistry Solving Problems A Chemistry Handbook Matter and Change

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Chemistry Decoded Leonard W. Fine, 1976

Chemistry Martin Stuart Silberberg, 1996 This general chemistry text centres on the theme that observable change in chemical systems is the result of molecular change The aims of this edition are to enable students to perceive matter and change at the molecular level and to help build student confidence in their ability to solve chemical problems as they discover the relevance of chemistry to their lives

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Modern Chemistry Raymond E. Davis, 1999 2000 2005 State Textbook Adoption Rowan Salisbury

How to Survive (and Even Excel In) General Chemistry Elizabeth Kean, Catherine Middlecamp, 1994 A different kind of book about chemistry which teaches readers the process of learning chemistry not the topic itself Proving a valuable supplement to any introductory text this guide offers inside information to help make chemistry less stressful even enjoyable Includes exercises and sections for self assessment

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