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J. J. Lagowski



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Principles of Inorganic Chemistry Brian W. Pfennig, 2022-02-02 PRINCIPLES OF INORGANIC CHEMISTRY Discover the foundational principles of inorganic chemistry with this intuitively organized new edition of a celebrated textbook In the newly revised Second Edition of Principles of Inorganic Chemistry experienced researcher and chemist Dr Brian W Pfennig delivers an accessible and engaging exploration of inorganic chemistry perfect for sophomore level students This redesigned book retains all of the rigor of the first edition but reorganizes it to assist readers with learning and retention In depth boxed sections include original mathematical derivations for more advanced students while topics like atomic and molecular term symbols symmetry coordinates in vibrational spectroscopy polyatomic MO theory band theory and Tanabe Sugano diagrams are all covered Readers will find many worked examples throughout the text as well as numerous unanswered problems at varying levels of difficulty Informative colorful illustrations also help to highlight and explain the concepts discussed within The new edition includes an increased emphasis on the comparison of the strengths and weaknesses of different chemical models the interconnectedness of valence bond theory and molecular orbital theory as well as a more thorough discussion of the atoms in molecules topological model Readers will also find A thorough introduction to and treatment of group theory with an emphasis on its applications to chemical bonding and spectroscopy A comprehensive exploration of chemical bonding that compares and contrasts the traditional classification of ionic covalent and metallic bonding In depth examinations of atomic and molecular orbitals and a nuanced discussion of the interrelationship between VBT MOT and band theory A section on the relationship between a molecule's structure and bonding and its chemical reactivity With its in depth boxed discussions this textbook is also ideal for senior undergraduate and first year graduate students in inorganic chemistry Principles of Inorganic Chemistry is a must have resource for anyone seeking a principles based approach with theoretical depth Furthermore it will be useful for students of physical chemistry materials science and chemical physics

Thermodynamics and Equilibria in Earth System Sciences: An Introduction Jack J. Middelburg, 2024-02-27 Thermodynamics is needed to understand many processes on Earth be they physical chemical or biological Thermodynamics is critical to study the atmosphere lapse rate fohn winds circulation hydrosphere latent and sensible heat pressure dependence of freezing boiling points geosphere geothermal gradients mineral stability and the biosphere redox zonation evolution of biogeochemical cycles This introduction to thermodynamics and equilibria aims to provide the basic concepts of relevance for atmospheric marine climate and environmental sciences and to prepare students for more advanced classes in physical chemistry mineralogy and petrology This is an open access book [Feyerabend's Epistemological Anarchism](#) Mansoor Niaz, 2020-01-27 This book argues that the traditional image of Feyerabend is erroneous and that contrary to common belief he was a great admirer of science It shows how Feyerabend presented a vision of science that represented how science really works Besides giving a theoretical framework based on Feyerabend's philosophy of science the book

offers criteria that can help readers to evaluate and understand research reported in important international science education journals with respect to Feyerabend's epistemological anarchism. The book includes an evaluation of general chemistry and physics textbooks. Most science curricula and textbooks provide the following advice to students: Do not allow theories in contradiction with observations and all scientific theories must be formulated inductively based on experimental facts. Feyerabend questioned this widely prevalent premise of science education in most parts of the world and in contrast gave the following advice: Scientists can accept a hypothesis despite experimental evidence to the contrary and scientific theories are not always consistent with all the experimental data. No wonder Feyerabend became a controversial philosopher and was considered to be against rationalism and anti-science. Recent research in philosophy of science, however, has shown that most of Feyerabend's philosophical ideas are in agreement with recent trends in the 21st century. Of the 120 articles from science education journals evaluated in this book, only 9% recognized that Feyerabend was presenting a plurality of perspectives based on how science really works. Furthermore, it has been shown that Feyerabend could even be considered as a perspectival realist. Among other aspects, Feyerabend emphasized that in order to look for breakthroughs in science, one does not have to be complacent about the truth of the theories but rather has to look for opportunities to break rules or violate categories. Mansoor Niaz carefully analyses references to Feyerabend in the literature and displays the importance of Feyerabend's philosophy in analyzing historical episodes. Niaz shows through this remarkable book a deep understanding to the essence of science. Calvin Kalman, Concordia University, Canada. In this book, Mansoor Niaz explores the antecedents, context, and features of Feyerabend's work and offers a more nuanced understanding, then reviews and considers its reception in the science education and philosophy of science literature. This is a valuable contribution to scholarship about Feyerabend with the potential to inform further research as well as science education practice. David Geelan Griffith, University of Australia.

[Applied Chemistry](#) Siddharth Venkatesh, 2025-02-20. Discover the essential aspects of chemistry in various industries with [Applied Chemistry: Practical Applications](#). This comprehensive textbook provides an in-depth understanding of fundamental chemical principles and their real-world applications. Covering a wide range of topics from chemical reactions and materials science to environmental chemistry and sustainable practices, it caters to students, researchers, and professionals. Written by experts, our book blends theoretical concepts with practical examples, offering a solid foundation in key concepts followed by discussions on their applications in industry, technology, and everyday life. We emphasize sustainability, green chemistry principles, and environmentally friendly practices. Clear explanations of complex topics are supported by diagrams, illustrations, and tables. Our book integrates modern research findings and technological advancements in chemistry. End-of-chapter summaries, review questions, and exercises reinforce learning and facilitate self-assessment. Supplementary materials, including online resources and laboratory exercises, enhance the learning experience. Whether you're a student seeking an introduction to applied chemistry or a professional looking to expand your knowledge,

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Evolving Nature of Objectivity in the History of Science and its Implications for Science Education Mansoor Niaz, 2017-10-26 This book explores the evolving nature of objectivity in the history of science and its implications for science education It is generally considered that objectivity certainty truth universality the scientific method and the accumulation of experimental data characterize both science and science education Such universal values associated with science may be challenged while studying controversies in their original historical context The scientific enterprise is not characterized by objectivity or the scientific method but rather controversies alternative interpretations of data ambiguity and uncertainty Although objectivity is not synonymous with truth or certainty it has eclipsed other epistemic virtues and to be objective is often used as a synonym for scientific Recent scholarship in history and philosophy of science has shown that it is not the experimental data Baconian orgy of quantification but rather the diversity plurality in a scientific discipline that contributes toward understanding objectivity History of science shows that objectivity and subjectivity can be considered as the two poles of a continuum and this dualism leads to a conflict in understanding the evolving nature of objectivity The history of objectivity is nothing less than the history of science itself and the evolving and varying forms of objectivity does not mean that one replaced the other in a sequence but rather each form supplements the others This book is remarkable for its insistence that the philosophy of science and in particular that discipline s analysis of objectivity as the supposed hallmark of the scientific method is of direct value to teachers of science Meticulously yet in a most readable way Mansoor Niaz looks at the way objectivity has been dealt with over the years in influential educational journals and in textbooks it s fascinating how certain perspectives fade while basic questions show no sign of going away There are few books that take both philosophy and education seriously this one does Roald Hoffmann Cornell University chemist writer and Nobel Laureate in Chemistry

Stantec's Water Treatment John C. Crittenden, R. Rhodes Trussell, David W. Hand, Kerry J. Howe, George Tchobanoglous, 2022-11-08 The updated third edition of the definitive guide to water treatment engineering now with all new online content Stantec s Water Treatment Principles and Design provides comprehensive coverage of the principles theory and practice of water treatment engineering Written by world renowned experts in the field of public water supply this authoritative volume covers all key aspects of water treatment engineering including plant design water chemistry and microbiology water filtration and disinfection residuals management internal corrosion of water conduits regulatory

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American Book Publishing Record, 1997-09 *Subject Guide to Books in Print*, 1997 **Macmillan Encyclopedia of Chemistry** J. J. Lagowski, 1997 Comprehensive and up to date this unique four volume set offers readers a complete overview of the broad spectrum of general chemistry It enables them to obtain a basic yet thorough understanding of matter the processes it undergoes the principles that govern it and the international cast of men and women who have been critical in the development of the science of chemistry From elements atoms and molecules to thermochemistry spectroscopy and chemical bonding its clear and concise explanations provide an illuminating and readily comprehensible introduction Key presentations include forty element definition articles each providing basic periodic table information and general information on the element in question Ninety five biographical articles deal with prominent chemists while other articles provide additional historical context particularly with respect to eighteenth nineteenth and twentieth century developments

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