

## Section-A ..... MCQs XI Chemistry (17 Marks)

Q.1 Select the most suitable answer from the given ones in each question:

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- Salt which is formed by the neutralization of weak acid and strong base is:  
(a)  $\text{NaNO}_3$  (b)  $\text{NH}_4\text{Cl}$  (c)  $\text{Na}_2\text{CO}_3$  (d)  $\text{NH}_4\text{CN}$
- Which one of the following cannot have value equal to zero?  
(a) Activation energy (b) Reduction potential (c) Principal quantum number (d) Both a and c
- If the volume occupied by oxygen gas ( $\text{O}_2$ ) at STP is  $44.8 \text{ dm}^3$ , the number molecules of  $\text{O}_2$  in the vessels are:  
(a)  $3.01 \times 10^{23}$  (b)  $6.02 \times 10^{23}$  (c)  $12.04 \times 10^{23}$  (d)  $24.08 \times 10^{23}$
- The number of carbon atoms in half mole of sugar ( $\text{C}_{12}\text{H}_{22}\text{O}_{11}$ ) are approximately:  
(a)  $6 \times 10^{23}$  (b)  $36 \times 10^{23}$  (c)  $60 \times 10^{23}$  (d)  $72 \times 10^{23}$
- The strongest oxidizing agent and strongest reducing agent in the electro chemical series respectively is:  
(a) Li and  $\text{H}_2$  (b) Li and  $\text{H}_2$  (c) F and Li (d)  $\text{H}_2$  and F
- Least entropy found in which of the following state of water:  
(a) Steam at  $100^\circ\text{C}$  (b) Liquid water at  $25^\circ\text{C}$  (c) Liquid water at  $4^\circ\text{C}$  (d) Ice at  $0^\circ\text{C}$
- Which of the following possesses weakest London dispersion forces:  
(a)  $\text{Cl}_2$  (b)  $\text{F}_2$  (c)  $\text{Br}_2$  (d)  $\text{I}_2$
- In NaCl, each Na ion is surrounded by Cl ions in the numbers:  
(a) Four (b) Three (c) Six (d) Seven
- Oxidation number of Fe in  $\text{Fe}_3\text{O}_4$  is: (Extra MCQ)  
(a)  $+3/8$  (b)  $+8/3$  (c)  $+4/3$  (d)  $+3/2$
- This molecule has zero dipole moment  
(a)  $\text{CCl}_4$  (b)  $\text{SO}_2$  (c)  $\text{CO}_2$  (d) Both a and c
- The number of bonds in ethyne ( $\text{C}_2\text{H}_2$ ) is:  
(a) 2 sigma, 2 pi (b) 3 sigma, two pi (c) 3 sigma, 3 pi (d) 2 sigma, 3 pi
- The potential energy of an electron can be denoted by  
(a)  $\frac{Ze^2}{8\pi\epsilon_0 r}$  (b)  $-\frac{Ze^2}{4\pi\epsilon_0 r}$  (c)  $-\frac{Ze^2}{8\pi\epsilon_0 r}$  (d) None of them
- A colloidal solution of liquid into liquid is known as:  
(a) Gel (b) Foam (c) Sol (d) Emulsion
- Which of the following does NOT alter the pH of a solution?  
(a)  $\text{NH}_4\text{Cl}$  (b)  $\text{Na}_2\text{CO}_3$  (c)  $\text{NaCl}$  (d)  $\text{Mg}(\text{OH})\text{Cl}$
- The unit of rate constant (K) for the first order reaction is:  
(a)  $\text{s}^{-1}$  (b)  $\text{conc. s}^{-1}$  (c)  $\text{conc}^{-1} \text{ s}$  (d)  $\text{conc}^{-1} \text{ s}^{-1}$
- The decomposition of  $\text{H}_2\text{O}_2$  is inhibited by:  
(a) 2% ethanol (b) Glycerin (c)  $\text{MnO}_2$  (d)  $\text{V}_2\text{O}_5$
- The sum of mole fractions of components of a solution is equal to:  
(a) 0.0 (b) 1.0 (c) 10 (d) 100
- The conduction of electricity through an electrolytic solution is due to the flow of  
(a) Electrons (b) ions (c) Atoms (d) Molecules

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