

## Acids & Bases Review

1. Name the following compounds as acids and circle which are weak acids:



nitric acid



sulfuric acid



hydrofluoric acid



carbonic acid



acetic acid



phosphoric acid

2. Write the formulas for the following bases:

a) calcium hydroxide



b) aluminum hydroxide



c) ammonium hydroxide



d) lithium hydroxide



3. Complete the following table:

	Acids	Bases
Definition	A substance that increases the concentration of $\text{H}^+$ (or $\text{H}_3\text{O}^+$ ) when added to water. or A substance that acts as a proton ( $\text{H}^+$ ) donor.	A substance that increases the concentration of $\text{OH}^-$ when added to water. or A substance that acts as a proton ( $\text{H}^+$ ) acceptor.
Electrolyte in aqueous solution?	$\text{H}^+$	$\text{OH}^-$
pH	less than 7	greater than 7
Taste	sour	bitter
Litmus turns	blue turns red	red turns blue
Which reacts with metals to form $\text{H}_2$ gas?	yes	no

**Conjugate acid-base pairs:**

4. Identify the acid, the base, the conjugate acid and the conjugate base in the reactions:



5. An acid will form the conjugate base and a base will form the conjugate acid.

# Chemistry Acids And Bases D Answers

**Jessica J Manson**



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web transfer of pollen grains from the male part of the flower called the stamen to the female part of the flower called the



pistil this fertilizes the female flower and enables it to produce seed and fruit

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