

1 6. Which condition is necessary for most gases to behave nearly ideally?

- a. high pressure
- b. high temperature
- c. low compressibility
- d. low expansion

7. Which of the following is *not* a physical property of gases?

- a. absence of definite volume
- b. large density
- c. high compressibility
- d. fluidity

8. Rates of effusion of different gases are proportional to their

- a. polarity.
- b. particle charge.
- c. particle velocities.
- d. compressibility.

9. Which states of matter are fluid?

- a. gases and liquids
- b. liquids and solids
- c. gases only
- d. liquids only

10. Which gas deviates the most from ideal behavior?

- a.  $\text{CH}_4$
- b.  $\text{H}_2\text{O}$
- c.  $\text{Cl}_2$
- d. He

# Chemistry Ch 10 Section 1 Review Answers

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