

## Chemistry Worksheets Class 10 on Chapter 1 Chemical Reactions and Equations- Set 1

**Q-1:** Write the balanced chemical equations for each of the following reactions.

- a) Liquid hydrogen peroxide decomposes to form water and oxygen gas.
- b) Acetylene gas burns in oxygen to form carbon dioxide and water along with evolution of heat.
- c) When solid mercury(II) oxide is heated, liquid mercury and oxygen gas are produced.
- d) Marble dissolves in a hydrochloric acid to give calcium chloride, water and carbon dioxide.
- e) Copper sulphate on treatment with potassium iodide precipitates cuprous iodide, liberates iodine gas and also forms potassium sulphate.

**Q-2:** Fill in the blanks

- a) \_\_\_\_\_ is the magnetic oxide of iron.
- b) In reactivity series, manganese lies \_\_\_\_\_ zinc
- c) To the reaction between  $\text{AgNO}_3$  solution and  $\text{NaCl}$  solution, the names given to this type of reactions are \_\_\_\_\_ and \_\_\_\_\_.
- d) Proteins decompose to form \_\_\_\_\_ in our body.
- e) If copper turnings are added to colourless  $\text{AgNO}_3$  solution, it turns \_\_\_\_\_ after sometime.

**Q-3:** Which of the following displacement reactions will not take place?

- a)  $\text{Zn(s)} + \text{FeSO}_4(\text{aq}) \rightarrow \text{ZnSO}_4(\text{aq}) + \text{Fe(s)}$
- b)  $\text{Fe(s)} + \text{CuSO}_4(\text{aq}) \rightarrow \text{FeSO}_4(\text{aq}) + \text{Cu(s)}$
- c)  $\text{Cu(s)} + 2\text{AgNO}_3(\text{aq}) \rightarrow \text{Cu(NO}_3)_2(\text{aq}) + 2\text{Ag(s)}$
- d)  $\text{Cu(s)} + \text{FeSO}_4(\text{aq}) \rightarrow \text{CuSO}_4(\text{aq}) + \text{Fe(s)}$

**Q-4:** Which of the following statements is false?

- a) Oxidation results in the loss of an electropositive element and the gain of an electronegative element.
- b) Reduction results in the loss of an electronegative element and the gain of an electropositive element.
- c) A species that gains electrons is referred to as a reducing agent.
- d) An oxidising agent is a type of specie that absorbs hydrogen atoms.

**Q-5:** Why are certain chemicals and medications kept in coloured bottles?

**Q-6:** Sakshi desired that her home be white washed. She went to the market and bought 20 kg of quick lime, which she dissolved in 60 litres of water. When she added lime to water, she noticed that the water began to boil even though it was not heated. Give an explanation for her observation. Name the product and write the corresponding chemical equation and word equation.

# Chemistry Chapter 1 Worksheet

**Holt Rinehart & Winston, Holt,  
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## Chemistry Chapter 1 Worksheet:

**Chemistry (Teacher Guide)** Dr. Dennis Englin, 2018-02-26 This book was created to help teachers as they instruct students through the Master's Class Chemistry course by Master Books. The teacher is one who guides students through the subject matter, helps each student stay on schedule and be organized, and is their source of accountability along the way. With that in mind, this guide provides additional help through the laboratory exercises as well as lessons, quizzes, and examinations that are provided along with the answers. The lessons in this study emphasize working through procedures and problem solving by learning patterns. The vocabulary is kept at the essential level. Practice exercises are given with their answers so that the patterns can be used in problem solving. These lessons and laboratory exercises are the result of over 30 years of teaching home school high school students and then working with them as they proceed through college. Guided labs are provided to enhance instruction of weekly lessons. There are many principles and truths given to us in Scripture by the God that created the universe and all of the laws by which it functions. It is important to see the hand of God and His principles and wisdom as it plays out in chemistry. This course integrates what God has told us in the context of this study.

**Features**  
Each suggested weekly schedule has five easy-to-manage lessons that combine reading and worksheets. Worksheets, quizzes, and tests are perforated, and three-hole punched materials are easy to tear out, hand out, grade, and store. Adjust the schedule and materials needed to best work within your educational program. Space is given for assignments, dates. There is flexibility in scheduling. Adapt the days to your school schedule.

**Workflow**  
Students will read the pages in their book and then complete each section of the teacher guide. They should be encouraged to complete as many of the activities and projects as possible as well. Tests are given at regular intervals with space to record each grade.

**About the Author**  
DR DENNIS ENGLIN earned his bachelor's from Westmont College, his master of science from California State University, and his EdD from the University of Southern California. He enjoys teaching animal biology, vertebrate biology, wildlife biology, organismic biology, and astronomy at The Master's University. His professional memberships include the Creation Research Society, the American Fisheries Association, Southern California Academy of Sciences, Yellowstone Association, and Au Sable Institute of Environmental Studies.

*Survey of Science History & Concepts Parent Lesson Plan*, 2013-08-01

**Survey of Science History Concepts Course Description**  
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**Semester 1 Math and Physics**  
Numbers surround us. Just try to make it through a day without using any. It's impossible: telephone numbers, calendars, volume settings, shoe sizes, speed limits, weights, street numbers, microwave timers, TV channels, and the list goes on and on. The many advancements and branches of mathematics were developed through the centuries as people encountered problems and relied upon math to solve them. It's amazing how ten simple digits can be used in an

endless number of ways to benefit man The development of these ten digits and their many uses is the fascinating story in Exploring the World of Mathematics Physics is a branch of science that many people consider to be too complicated to understand John Hudson Tiner puts this myth to rest as he explains the fascinating world of physics in a way that students can comprehend Did you know that a feather and a lump of lead will fall at the same rate in a vacuum Learn about the history of physics from Aristotle to Galileo to Isaac Newton to the latest advances Discover how the laws of motion and gravity affect everything from the normal activities of everyday life to launching rockets into space Learn about the effects of inertia first hand during fun and informative experiments Exploring the World of Physics is a great tool for student who want to have a deeper understanding of the important and interesting ways that physics affects our lives Semester 2 Biology and Chemistry The field of biology focuses on living things from the smallest microscopic protozoa to the largest mammal In this book you will read and explore the life of plants insects spiders and other arachnids life in water reptiles birds and mammals highlighting God s amazing creation You will learn about biological classification how seeds spread around the world long term storage of energy how biologists learned how the stomach digested food the plant that gave George de Mestral the idea of Velcro and so much more For most of history biologists used the visible appearance of plants or animals to classify them They grouped plants or animals with similar looking features into families Starting in the 1990 s biologists have extracted DNA and RNA from cells as a guide to how plants or animals should be grouped Like visual structures these reveal the underlying design of creation Exploring the World of Biology is a fascinating look at life from the smallest proteins and spores to the complex life systems of humans and animals Chemistry is an amazing branch of science that affects us every day yet few people realize it or even give it much thought Without chemistry there would be nothing made of plastic there would be no rubber tires no tin cans no televisions no microwave ovens or something as simple as wax paper This book presents an exciting and intriguing tour through the realm of chemistry as each chapter unfolds with facts and stories about the discoveries of discoverers Find out why pure gold is not used for jewelry or coins Join Humphry Davy as he made many chemical discoveries and learn how they shortened his life See how people in the 1870s could jump over the top of the Washington Monument Exploring the World of Chemistry brings science to life and is a wonderful learning tool with many illustrations and biographical information *Prentice Hall Science Explorer* Michael J. Padilla,2002 **Engineering with Mathcad** Brent Maxfield,2006-11-18 Using the author s considerable experience of applying Mathcad to engineering problems Engineering with Mathcad identifies the most powerful functions and features of the software and teaches how to apply these to create comprehensive engineering calculations Many examples from a variety of engineering fields demonstrate the power and utility of Mathcad s tools while also demonstrating how other software such as Microsoft Excel spreadsheets can be incorporated effectively This simple step by step approach makes this book an ideal Mathcad text for professional engineers as well as engineering and science students A CD ROM packaged with the book contains all the

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