

# **CHEMISTRY MODULES OBJECTIVE AND SYLLABUS**

## **LEVEL 1 MODULES**

### **CM1111: Inorganic Chemistry 1 (Essential Module)**

**Workload: 4-1-0-2-3**

#### **Aims & Objectives**

This module introduces some basic concepts of modern inorganic chemistry. In the first part, the atomic structure, the bonding models (Lewis structure, valence bond theory, basic molecular theory), basic concepts of acids and bases, redox reactions will be covered. In the part 2, the periodicity and chemistry of most main group elements will be included.

#### **Syllabus**

##### **Part One: Basic Concepts**

1. Atomic Structure
  - 1) Atomic Quantization
  - 2) Schrödinger Equation for hydrogen atom
  - 3) Many-electron systems
2. Chemical periodicity and the periodic Table
3. Molecular structure and bonding
  - 1) Lewis structure
  - 2) Valence bond theory
  - 3) Basic molecular orbital theory
4. Acid - Base and donor - acceptor chemistry
  - 1) Brønsted-Lowry concept
  - 2) Lewis concept
5. Reduction and Oxidation

##### **Part Two: Chemistry of Selected Main Group Elements**

1. Hydrogen and Group 1
2. Group 2
3. Group 13
4. Group 14
5. Group 15
6. Group 16
7. Group 17
8. Group 18

### **CM1121: Organic Chemistry 1 (Essential Module)**

**Workload: 4-2-0-2-2**

#### **Aims & Objectives**

The module deals primarily with the basic principles to understand the structure and reactivity of organic molecules. Emphasis is on substitution and elimination reactions and chemistry of various functional groups. You will be taught the basic concepts on how simple molecules can be constructed. Reactions mechanism, organic transformations and stereochemistry will also be discussed.

By reading this modules, students should be able to:

- i. Understand and be able to apply and evaluate simple organic reaction transformations, functional group interconversion and C-C bond formation reactions.
- ii. Understand the scope and limitations as well as the mechanisms of organic reactions.
- iii. Understand the importance of stereochemistry and be able to view three-dimensional structures with ease.
- iv. Describe and explain currently held views of chemical bonding, and account for the chemical reactivity of reactions.

# Chemistry Modules Objective And Syllabus

**D Keegan**



## **Chemistry Modules Objective And Syllabus:**

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**Chinese Science Education in the 21st Century: Policy, Practice, and Research** Ling L. Liang,Xiufeng Liu,Gavin W. Fulmer,2016-08-16 This book provides an overview of science education policies research and practices in mainland China with specific examples of the most recent developments in these areas It presents an insiders report on the status of Chinese science education written primarily by native speakers with first hand experiences inside the country In addition the book features multiple sectional commentaries by experts in the field that further connect these stories to the existing science education literature outside of China This book informs the international community about the current status of Chinese science education reforms It helps readers understand one of the largest science education systems in the world which includes according to the Programme for International Student Assessment the best performing economy in the world in science math and reading Shanghai China Readers gain insight into how science education in the rest of China compares to that in Shanghai the ways Chinese science educators teachers and students achieve what has been accomplished what Chinese students and teachers actually do inside their classrooms what educational policies have been helpful in promoting student learning what lessons can be shared within the international science education community and much more This book appeals to science education researchers comparative education researchers science educators graduate students state science education leaders and officers in the international communities It also helps Chinese students and faculty of science education discover effective ways to share their science education stories with the rest of the world      **Book of Abstracts** ,1977      **The Directory of Graduate Studies** ,1998      **Republic of Ireland** Peter E. Childs,1994      *Current Index to Journals in Education* ,1984      Journal of Engineering Education ,2002      **Documentation Abstracts** ,1988      **Graduate Studies** ,1994      **GCSE Science** Graham Hill,1997 Intended for achieving examination success this series replaces the former GCSE Study Guides This book which are part of the new Revise GCSE series offers complete study and reference support for the Science GCSE course as well as being a revision guide for the Science GCSE examinations and Scottish Standard Grade It provides a breakdown of the syllabuses and exam structure for every exam board advice on studying revising coursework and exams topic by topic coverage clearly presented with many examples and diagrams quick tests for every topic to check progress and a bank of recent GCSE exam questions with answers to fine tune exam techniques

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letter from the very beginning he plunges into the problems of the church c paul is deeply engaged in very practical matters ii paul s apostleship was called into question so he has to defend it iii

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web at corinth on the contrary he was not universally admired the spirit people at least were against him and on reflection in this moment of crisis he may have realized that the strategy he had adopted in 1 corinthians would not have made them any friendlier

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web the corinthians launched an invasion of corcyra herself meeting the enemy forces near the sybota islands at the mouth of the corfu channel this time the corinthians bolstered by peloponnesian allies were numerically superior to their foes and managed to rout the corcyraean forces in the initial engagement which prompted the athenians to

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