

## CHEMISTRY REPLACEMENT REACTION WORKSHEET

### DESCRIPTION

During double replacement, the cations and anions of two different compounds switch places.



### REACTION GUIDELINE

1. It is important that the formulas of the products be written correctly. If they are correct, balancing the equation is a simple task. If not, the equation will probably never balance.
2. In these reactions, there is never a change in oxidation state (in other words, the charges stay the same).

### EXAMPLES



Silver nitrate + sodium chloride  $\rightarrow$  silver chloride + sodium nitrate



Lead (II) nitrate + copper (II) sulfate  $\rightarrow$  lead (II) sulfate + copper (II) nitrate

### DOUBLE REPLACEMENT PRACTICE REACTIONS

For each reaction, predict the products and balance the equation. State the reaction in chemical formulas and in symbols. (For example:



Silver nitrate + sodium chloride  $\rightarrow$  silver chloride + sodium nitrate



# Chemistry Replacement Reaction Chem 121 Answers

**D Kirk**



## Chemistry Replacement Reaction Chem 121 Answers:

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Chemistry: The Study of Matter Henry Dorin, 1982

**UPCO's Review of Chemistry** Robert M. Capie, 2001

Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction

How to Prepare for the AP Chemistry Neil D. Jespersen, Neil Jespersen, PH. D., 1999-02 This manual offers two diagnostic exams to help students pinpoint their strengths and weaknesses plus three full length practice exams closely modeled on actual AP chemistry exams

**Chemical Abstracts**, 1914

Environmental Health Perspectives, 1993

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Fundamental Experiments for College Chemistry Harper Wine Frantz, Lloyd E. Malm, 1963

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**Engineering and Mining Journal**, 1926

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