

Questions

Q1.

Answer the question with a cross in the box you think is correct ☒ . If you change your mind about an answer, put a line through the box ☒ and then mark your new answer with a cross ☒ .

The halogens are elements in Group 7 of the Periodic Table.

The halogens can be identified by their colour in an organic solvent such as hexane or cyclohexane.

Which sequence of colours is correct for chlorine, bromine and iodine dissolved in an organic solvent?

(1)

	Chlorine	Bromine	Iodine
<input type="checkbox"/> A	orange	red-brown	black
<input type="checkbox"/> B	pale green	orange	black
<input type="checkbox"/> C	orange	red-brown	purple
<input type="checkbox"/> D	pale green	orange	purple

(Total for question = 1 mark)

Q2.

This question is about the elements in Group 7 of the Periodic Table and some of their compounds.

What is the colour of iodine in the solid and gas states?

(1)

	Colour of iodine solid	Colour of iodine gas
<input type="checkbox"/> A	purple	brown
<input type="checkbox"/> B	purple	purple
<input type="checkbox"/> C	grey/black	brown
<input type="checkbox"/> D	grey/black	purple

(Total for question = 1 mark)

Inorganic Chemistry Mcq Questions With Answers

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