

#### Covalent Bonding: Revision Questions

#### Bronze:

1. What is a covalent bond?

A shared pair of electrons (at least one pair) between two or more non-metal atoms.

2. What do covalent bonds form between?

Covalent bonds form between non-metals.

Right the formula for the following molecules: nitrogen, oxygen, water, chlorine, carbon dioxide and hydrogen.

No On Histor CT, CO, and His

4. How many carbon atoms make up Buckminsterfullerene?

60 carbon atoms (Carl

5. Draw dot-cross diagrams for: hydrogen chloride, ammonia and chlorine.







 Diamond and graphite made from which are both giant covalent structures that are element?

Corbon

#### Sibeen

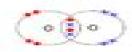
1. Fullerenes in the name used to describe what?

Hollow shaped molecules of carbon. The structure is based on hexagonal rings of carbon atoms.

- How many bonds does each carbon atom in a) diamond and b) graphite form?
- a) 4 bonds b) 3 bonds
  - 3. Draw dot cross diagrams for: methane, oxygen and nitrogen.



emergy to break





- 4. Why does diamond have such a high melting point?
  Each carbon atom makes 4 strong covalent bonds that requires a large amount of
- State some of the potential uses of graphene, linking to its structure.
   Graphene is a single sheet/layer of graphite. Graphene is an excellent conductor of heat and electricity. This makes it suitable for its use in solid support computer chips



# **Covalent Bonding Core Teaching Resources Test Answer**

Xiaolong Qi

#### **Covalent Bonding Core Teaching Resources Test Answer:**

Resources for Teaching Middle School Science Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents Chemistry in the Community ,1995 **Proceedings of the National Convention** National Science Teachers Association, National Science Teachers Association. A report of the proceedings [of the] national convention, 1959 <u>Teaching Chemical Bonding</u> Margaret Irene Lindsay,1995 This document presents an instructional strategy for teaching chemical bonding using parables and music Games student interactions and worksheets are included in the lesson plans

Topics include metallic bonding covalent bonding including molecular and network structure and ionic bonding JRH

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