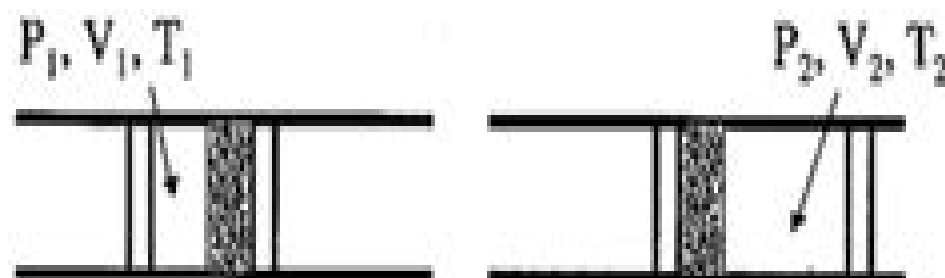


4. The connection between the temperature and pressure of a gas in the process just considered (the so-called Joule-Thomson process) is best expressed in terms of a thermodynamic derivative $j \equiv (\partial T / \partial P)_H$. Show that

$$j \equiv \left(\frac{\partial T}{\partial P} \right)_H = \frac{V}{C_P} (T\alpha - 1)$$

where α is the thermal expansion coefficient, and C_P is the heat capacity at constant pressure.

5. For the media described by the equation of state given in 2., , investigate how the sign of j depends on the parameters A and n_0 and state variables. State the conditions for $j < 0$, $j = 0$, and $j > 0$.



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