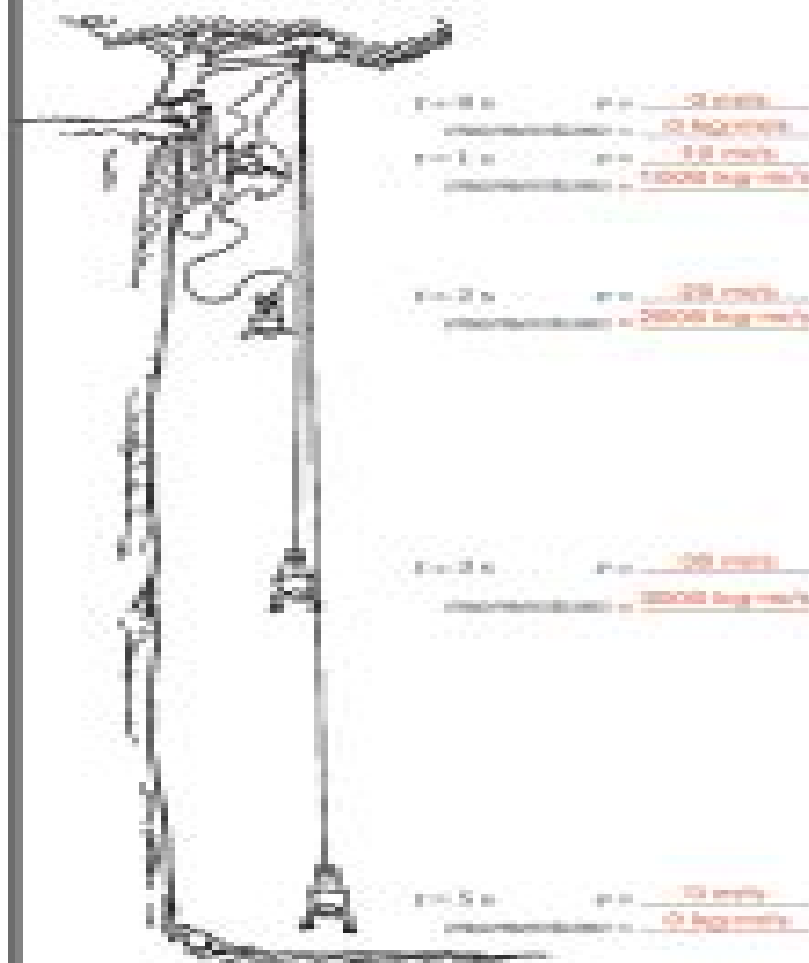


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9-3

Stowaways and Energy



Before Emma starts to get F1=Jerk to decrease and try bungee jumping, Gervais keeps there a high cliff and experiences freefall for 2 seconds. Then the bungee cord begins to stretch, reducing his speed to zero in 3 seconds. Fortunately, the cord stretches to its maximum length, just above of the ground below.

F1 is the Jerk. Emma's mass is 100 kg. Acceleration of free fall is 9.8 m/s^2 .

Express values in SI units, reference to m, velocity in m/s, acceleration in kg m/s², impulse in N-s, and deceleration in m/s².

What is the free-fall distance of Emma just before the bungee cord begins to stretch?

$y = 19.6 \text{ m}$

Area during the 2-s interval of free fall

$a = 9.8 \text{ kg-m/s}^2$

Area during the 3-s interval of slowing down

$a = 9.8 \text{ kg-m/s}^2$

Impulse during the 3-s interval of slowing down

$a = 9.8 \text{ N-s}$

Average force exerted by the cord during the 3-s interval of slowing down

$a = 9.8 \text{ N}$

How about work and energy? How much KE does Emma have 3 s after his jump?

$KE = 1960 \text{ J}$

How much does gravitational PE decrease during this 3-s?

$PE = 1960 \text{ J}$

What two kinds of PE are changing during the slowing-down interval?

Gravitational and elastic potential energies

CONCEPTUAL PHYSICS

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Metallurgical Society of AIME.



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CRC Handbook of Chemistry and Physics, 85th Edition David R. Lide, 2004-06-29 Get a FREE first edition facsimile with each copy of the 85th Researchers around the world depend upon having access to authoritative up to date data And for more than 90 years they have relied on the CRC Handbook of Chemistry and Physics for that data This year is no exception New tables extensive updates and added sections mean the Handbook has again set a new standard for reliability utility and thoroughness This edition features a Foreword by world renowned neurologist and author Oliver Sacks a free facsimile of the 1913 first edition of the Handbook and thumb tabs that make it easier to locate particular data New tables in this edition include Index of Refraction of Inorganic Crystals Upper and Lower Azeotropic Data for Binary Mixtures Critical Solution Temperatures of Polymer Solutions Density of Solvents as a Function of Temperature By popular request several tables omitted from recent editions are back including Coefficients of Friction and Miscibility of Organic Solvents Ten other sections have been substantially revised with some such as the Table of the Isotopes and Thermal Conductivity of Liquids significantly

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